

Original paper

WinGrt, a Windows program for garnet supergroup minerals

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A Microsoft® Visual Basic software, called WinGrt, has been developed to calculate and classify wet-chemical and electron-microprobe garnet supergroup mineral analyses. The program evaluates 33 approved species that belong to the tetragonal henritermierite and isometric bitikleite, schorlomite, garnet and berzeliite groups based on the Commission on New Minerals and Mineral Names (CNMMN) of the International Mineralogical Association (IMA–13) nomenclature scheme. WinGrt also evaluates thirty geothermometers using the Fe²⁺–Mg exchange reactions for garnet–biotite, garnet–clinopyroxene and garnet–orthopyroxene pairs within the application range of greenschist-, amphibolite-, granulite- and eclogite-facies metamorphic rocks. As naturally occurring garnet is potentially a useful provenance indicator, the program calculates end-member molecules from chemical compositions on the basis of different approaches and yields pyrope, almandine, spessartine, grossular, andradite and schorlomite phase on various ternary discrimination diagrams used in provenance studies. The ferric and ferrous iron contents from total FeO (wt. %) amount are estimated by stoichiometric constraints. The program allows the users to enter 30 input variables including Sample No, SiO₂, TiO₂, ZrO₂, HfO₂, Th₂O, SnO₂, Al₂O₃, Cr₂O₃, V₂O₃, Fe₂O₃, Mn₂O₃, Sc₂O₃, Y₂O₃ + REE₂O₃, FeO, MgO, MnO, ZnO, CaO, Na₂O, Li₂O, P₂O₅, V₂O₅, Sb₂O₅, As₂O₅, Nb₂O₅, UO₃, Te₂O₃, F and H₂O (wt. %). WinGrt also enables the user to enter the total REE₂O₃ (wt. %) as input values from La₂O₃ to Lu₂O₃ (wt. %) of garnet supergroup mineral analyses in program's data edit section. WinGrt enables the user to type or load multiple garnet compositions in the data entry section, to edit and load Microsoft® Excel files in calculating, classifying and naming the garnet species, and to store all the calculated parameters in the Microsoft® Excel file for further evaluation.

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1. Introduction

Garnet is one of the best-known minerals on the Earth and a key rock-forming mineral in crystalline rocks of the crust and mantle. The garnet minerals are present in diverse geologic settings and environments from igneous and metamorphic to sedimentary rocks, in remarkable crystal colors and size. In the Earth's crust, garnet is commonly found in metamorphic rocks derived from any protolith, from lower greenschist-facies rocks to ultrahigh-temperature granulites and ultrahigh-pressure eclogites (Baxter et al. 2013; Geiger 2013 and references therein). In igneous rocks, on the other hand, garnet commonly prefers to crystallize in peraluminous granites.

Garnet's density and its relative resistance to surface weathering and diagenetic processes lead to accumulation of this common detrital phase in the heavy mineral fraction of sediments. Consequently, certain garnet compositions including the major end-members such as almandine, pyrope, spessartine, and grossular have been empirically related to specific sources and, therefore, detrital garnets have been increasingly used in sedimentary provenance analysis. A number of binary and ternary discrimination diagrams for garnet have been proposed

in the literature (Mange and Morton 2007; Krippner et al. 2014; Suggate and Hall 2014; Tolosana-Delgado et al. 2018). In these diagrams, however, there is no exact and quantitative understanding of what garnet type originates from which host lithology.

Because of the flexible structure, at least 25 essential of total 53 elements were reported in the natural garnets (Geiger 2013; Grew et al. 2013). Garnet's high coordination numbers, as well as relatively high density, low compressibility and increasing stability with increasing pressure make it a petrologically and geochemically important mineral. According to the current IMA-approved nomenclature scheme (Grew et al. 2013), the garnet supergroup includes all minerals isostructural with garnet irrespective of which elements occupy the four atomic sites including three symmetry unique atomic sites (*X*, *Y*, *Z*) and the anionic site (ϕ). Hence, the general formula of the garnet supergroup minerals is expressed in the form of $\{X_3\}[Y_2](Z_3)\phi_{12}$, where *X*, *Y*, and *Z* refer to dodecahedral, octahedral and tetrahedral sites, respectively, and ϕ contains O²⁻, OH⁻, or F⁻ (Grew et al. 2013). Alternating $Z\phi_4$ tetrahedra and $Y\phi_6$ octahedra participate of corners to form a three-dimensional framework containing $X\phi_8$ triangular dodecahedra (Schingaro et al. 2016). In current

nomenclature of garnets (Grew et al. 2013), 33 species of the garnet supergroup were accepted, out of which thirty are classified into five groups on the basis of the symmetry and of the total charge of cations at the tetrahedral site. The standard garnet has a cubic symmetry, space group $Ia\bar{3}d$. However, two OH-bearing species, henritermierite and holtstamite, have tetragonal symmetry, space group $I4_1/acd$, with their X , Z , and ϕ sites split into more symmetrically unique atomic positions.

Chemistry of natural garnet group minerals plays an important role in provenance and geothermobarometry studies that require an accurate estimation of end-member components. Rickwood (1968) proposed a scheme for recasting garnet analyses into 19 end-member molecules with a discussion of the influence of calculation sequence on the end-members allocation. Rickwood's (1968) procedure of recasting garnet analyses into end-member molecules was later implemented by several computer programs and spreadsheet applications (Knowles 1987; Friberg 1989; Muhling and Griffin 1991; Yang 1991; Locock 2008). Although Rickwood's (1968) procedure has been widely used in literature, it was reviewed by Muhling and Griffin (1991) on the account of inappropriate treatment of titanium content. Consequently, a QuickBasic program was developed by Muhling and Griffin (1991) to produce 16 end-members that do not differ significantly from the Rickwood's (1968) procedure for most abundant garnets, but showing variations most notably in the Cr, Fe and Ti allocation. Locock (2008) developed an Excel spreadsheet to recast garnet analyses into 29 possible end-member components comprising 15 mineral species and 14 hypothetical end-members. Grew et al. (2013) proposed an IMA-approved nomenclature scheme for garnet supergroup minerals based on the dominant-valency rule and prepared an Excel spreadsheet to perform calculation, cation allocation, and classification of 32 approved species with group determination. Knierzinger et al. (2019) presented a new interactive MATLAB-based calculation and visualization tool, called TETGAR_C, to assess better spatial and quantitative differentiation trends of detrital garnet minerals including almandine–pyrope–grossular–spessartine end-members in the tetrahedral plot system for provenance studies.

In this paper, we present a new computer program, called WinGrt, developed using the Microsoft® Visual Basic programming language to calculate multiple garnets mineral data obtained by both wet-chemical and electron-microprobe techniques. The program recalculates garnet supergroup mineral analyses based on 8 cations and 12 anions. Calculation and classification of garnet supergroup mineral analyses are carried out according to the current IMA report (Grew et al. 2013). The program is capable of estimating the Fe^{3+} and Fe^{2+} contents from

microprobe-derived total FeO (wt. %) analysis using the stoichiometric constraints proposed by Droop (1987). Using the recalculated garnet analyses, as well as the entered biotite (Al^{VI} , Ti, Fe^{2+} , and Mg) and pyroxene (Al_{tot} , Fe_{tot} , Fe^{2+} , Mg) cations (*apfu*), together with input pressure (kbar) values, WinGrt implements thirty geothermometers for garnet–biotite, garnet–clinopyroxene, and garnet–orthopyroxene pairs. WinGrt allows the user to display garnet supergroup minerals in various ternary classification and variation diagrams by using the Golden Software's Grapher program. When compared to the previously published garnet-related computer programs and Excel spreadsheets, WinGrt provides the users a quick evaluation of multiple garnet analyses for classification as well as geothermometric calculations based on the Fe^{2+} –Mg exchange reactions between garnet–biotite, garnet–clinopyroxene and garnet–orthopyroxene pairs and sedimentary provenance analysis using garnet end-members in ternary classification diagrams.

2. Program description

Numerous Excel spreadsheets and executable computer programs have been developed and published for calculation and classification of rock-forming silicate group minerals including amphibole (Yavuz 2007; Locock 2014), mica (Yavuz 2003a, b), pyroxene (Yavuz 2001), garnet (Locock 2008; Grew et al. 2013), tourmaline (Yavuz et al. 2006, 2014), chlorite (Yavuz et al. 2015), epidote (Yavuz and Yıldırım 2018) in recent years. However, only a limited number of computer programs appeared in literature for silicate group mineral calculation, classification and thermobarometry estimation (e.g. Putirka 2008; Hora et al. 2013; Yavuz 2013; Lanari et al. 2014; Yavuz et al. 2015; Yavuz and Döner 2017).

WinGrt is a user-friendly compiled program package (14 MB) for garnet supergroup mineral analyses developed for personal computers running in the Microsoft® Windows operating system. The program calculates cations (*apfu*) from electron-microprobe and wet-chemical garnet analyses and classifies 33 garnet species belonging to five groups including the tetragonal henritermierite group and isometric bitikleite, garnet and berzeliite groups, as well as three species (i.e. katoite, cryolithionite, yafsoanite) that are single representatives of potential groups in which the Z site is vacant or occupied by monovalent or divalent cations (Tab. 1). The current version of program estimates temperatures using ten garnet–biotite (e.g. Ferry and Spear 1978; Hodges and Spear 1982; Perchuk and Lavrent'eva 1983; Bhattacharya et al. 1992), ten garnet–clinopyroxene (Ellis and Green 1979; Krogh 1988; Krogh–Ravna 2000; Nakamura 2009) and ten garnet–orthopyroxene (Raith et al. 1983; Sen and

Tab. 1 A list of the 33 approved species in the garnet supergroup (from Grew et al. 2013) recognized by the WinGrt program

NOT SPECIFIED GROUP (Z charge = 0–6)	HENRITERMIERITE GROUP (Z charge = 8)	BITIKLEITE GROUP (Z charge = 9)	SCHORLOMITE GROUP (Z charge = 10)	GARNET GROUP (Z charge = 12)	BERZELIITE GROUP (Z charge = 15)
Katoite $\text{Ca}_3\text{Al}_2\text{□}(\text{OH})_{12}$	Holtstamite $\text{Ca}_3\text{Al}_2\text{Si}_2\text{I□O}_8(\text{OH})_4$	Bitikleite $\text{Ca}_3\text{Sb}^{5+}\text{Sn}^{4+}\text{Al}_3\text{O}_{12}$	Kimzeyite $\text{Ca}_3\text{Zr}_2\text{SiAl}_2\text{O}_{12}$	Menzerite-(Y) $\text{Y}_2\text{CaMg}_2\text{Si}_3\text{O}_{12}$	Schäferite $\text{Ca}_2\text{NaMg}_2\text{V}^{5+}_3\text{O}_{12}$
Cryolithionite $\text{Na}_3\text{Al}_2\text{Li}_3\text{F}_{12}$	Henritermierite $\text{Ca}_3\text{Mn}^{3+}_2\text{Si}_2\text{I□O}_8(\text{OH})_4$	Usturite $\text{Ca}_3\text{Sb}^{5+}\text{ZrFe}^{3+}_3\text{O}_{12}$	Irinarassite $\text{Ca}_3\text{Sn}^{4+}_2\text{SiAl}_2\text{O}_{12}$	Pyrope $\text{Mg}_3\text{Al}_2\text{Si}_3\text{O}_{12}$	Palenzonaite $\text{Ca}_2\text{NaMn}^{2+}_2\text{V}^{5+}_3\text{O}_{12}$
Yafsoanite $\text{Ca}_3\text{Te}^{6+}_2\text{Zn}_3\text{O}_{12}$		Dzhuluite $\text{Ca}_3\text{Sb}^{5+}\text{Sn}^{4+}\text{Fe}^{3+}_3\text{O}_{12}$	Schorlomite $\text{Ca}_3\text{Ti}_2\text{SiFe}^{3+}_2\text{O}_{12}$	Grossular $\text{Ca}_3\text{Al}_2\text{Si}_3\text{O}_{12}$	Berzeliite $\text{Ca}_2\text{NaMg}_2\text{As}^{5+}_3\text{O}_{12}$
		Elbrusite $\text{Ca}_3\text{U}^{6+}_{0.5}\text{Zr}_{1.5}\text{Fe}^{3+}_3\text{O}_{12}$	Kerimasite $\text{Ca}_3\text{Zr}_2\text{SiFe}^{3+}_2\text{O}_{12}$	Spessartine $\text{Mn}^{2+}_3\text{Al}_2\text{Si}_3\text{O}_{12}$	Maganberzeliite $\text{Ca}_2\text{NaMn}^{2+}_2\text{As}^{5+}_3\text{O}_{12}$
			Toturite $\text{Ca}_3\text{Sn}^{4+}_2\text{SiFe}^{3+}_2\text{O}_{12}$	Almandine $\text{Fe}^{2+}_3\text{Al}_2\text{Si}_3\text{O}_{12}$	
			Hutcheonite† $\text{Ca}_3\text{Ti}_2(\text{SiAl})\text{O}_{12}$	Eringaite $\text{Ca}_3\text{Sc}_2\text{Si}_3\text{O}_{12}$	
				Goldmanite $\text{Ca}_3\text{V}^{3+}_2\text{Si}_3\text{O}_{12}$	
				Momoiite $\text{Mn}^{2+}_3\text{V}^{3+}_2\text{Si}_3\text{O}_{12}$	
				Knorringite $\text{Mg}_3\text{Cr}^{3+}_2\text{Si}_3\text{O}_{12}$	
				Uvarovite $\text{Ca}_3\text{Cr}^{3+}_2\text{Si}_3\text{O}_{12}$	
				Andradite $\text{Ca}_3\text{Fe}^{3+}_2\text{Si}_3\text{O}_{12}$	
				Calderite $\text{Mn}^{2+}_3\text{Fe}^{3+}_2\text{Si}_3\text{O}_{12}$	
				Majorite $\text{Mg}_3\text{SiMgSi}_3\text{O}_{12}$	
				Morimotoite $\text{Ca}_3\text{TiFe}^{2+}_2\text{Si}_3\text{O}_{12}$	

† New garnet species approved by the IMA (IMA 2013-029) later than the subcommittee report by Grew et al. (2013)

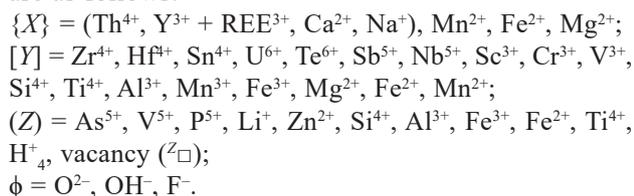
Bhattacharya 1984; Perchuk and Lavrent'eva 1990; Lal 1993) geothermometers based on the exchange reactions between Fe^{2+} and Mg.

Heavy minerals have been used in a number of provenance studies to discriminate and identify the source areas. Garnets are one of the commonly employed minerals due to their occurrence in a wide range of rocks, relative stability under both weathering and diagenetic conditions, as well as a wide range of major-element compositions. WinGrt calculates end-member molecules such as pyrope, almandine, spessartine, grossular, andradite, and schorlomite and allows the users to display them on various ternary classification diagrams in sedimentary provenance studies. A list of the calculation steps in the *Calculation Screen* and in an Excel output of the WinGrt is given in Tab. 2. Current version presents eight ternary garnet classification and nine garnet provenance-related ternary plots. These plots are displayed by the Golden Software's Grapher program by selecting diagram types from the pull-down menu of *Graph* in the *Calculation Screen* of WinGrt.

2.1. Site allocation of cations in the garnet supergroup and determination of garnet species

Classification of a rock-forming silicate mineral species is based on the chemical composition of an analyzed sample obtained by wet-chemical or electron-microprobe techniques. Proper site allocation of rock-forming minerals is an important issue, commonly demanding single-crystal, or Rietveld structure refinement using X-ray or neutron diffraction methods. Although most researchers have only electron-microprobe analyses, which provide no direct evidence of valence state, spectroscopic data are also required for unambiguous site assignment, especially when constituents could be present in more than one valence state. The general formula for the garnet supergroup minerals is expressed in the form of $\{X_3\}[Y_2](Z_3)\phi_{12}$, where X, Y, and Z refer to dodecahedral, octahedral, and tetrahedral sites, respectively, and ϕ is O, OH, or F (Grew et al. 2013). The site occupancies used in the

garnet supergroup general formula by WinGrt program are as follows:



Several elements that rarely exceed 1 wt. % in natural garnets (e.g. B, S, K, Ni, or Sr) as well as Ge, Ga and Pb that appear in significant amounts, but with rare occurrence, have been excluded from program's data entry and calculation steps. The following calculation and site allocation procedures were implemented in nomenclature of the garnet supergroup minerals by WinGrt program:

- (1) Calculation of formula from the chemical analysis is performed based on 8 cations and 12 anions. Ferric and ferrous iron separation from electron-microprobe analysis, as well as the proportion of Mn^{2+} and Mn^{3+} , are assessed using the Droop's (1987) method. If quantitative F (wt. %) and H_2O (wt. %) data are available, then ${}^Z\Box = 1/4F + 1/4H$.
- (2) Li^+ , Zn^{2+} , P^{5+} , As^{5+} and V^{5+} are allocated to the Z site. If $\text{Li}^+ < 1/4 F$, then sufficient vacancies are assumed to make up the deficit.
- (3) Si^{4+} is first allocated to the Z site. If $\text{Si} (apfu) > 3$ (including vacancies), then excess Si is added to the Y site.
- (4) Al^{3+} is first allocated to the Z site to bring total to 3 (*apfu*), then to the Y site.
- (5) Fe^{3+} is first allocated to the Z site to bring total to 3 (*apfu*), then to the Y site.
- (6) Ca^{2+} , Na^+ , Y^{3+} , REE^{3+} , and Th^{4+} are allocated to the X site.
- (7) Al^{3+} (remaining aluminum content at the Z site; i.e. $\text{Al}_{\text{tot}} - {}^Z\text{Al}$), Sc^{3+} , Ti^{4+} , V^{3+} , Cr^{3+} , Mn^{3+} , Fe^{3+} (remaining ferric iron content at the Z site; i.e. $\text{Fe}^{3+}_{\text{tot}} - {}^Z\text{Fe}^{3+}$), Zr^{4+} , Hf^{4+} , Nb^{5+} , Sn^{4+} , Sb^{5+} , Te^{6+} , and U^{6+} are allocated to the Y site. If the Z site is still $< 3 (apfu)$, then Fe^{2+} is added to bring the Z site total to 3 (*apfu*). If the content of Y site exceeds 2 (*apfu*), and the Z site is $< 3 (apfu)$, then Ti^{4+} is moved to the Z site to bring the Z site total to 3 (*apfu*).
- (8) Mg^{2+} is first allocated to the Y site to bring total to 2 (*apfu*), then remaining to the X site.
- (9) Fe^{2+} (remaining ferrous iron content at the Z site; i.e. $\text{Fe}^{2+}_{\text{tot}} - {}^Z\text{Fe}^{2+}$) is first allocated to the Y site to bring total to 2 (*apfu*), then to the X site.
- (10) Mn^{2+} is first allocated to the Y site to bring total to 2 (*apfu*), then to the X site.

According to Grew et al. (2013), the site allocation procedure listed above fails to differentiate holtstamite, a member of the hentermierite group, from grossular of

the garnet group, which would require additional information such as optical properties or crystallographic data. Following the classification procedures by Henry et al. (2011) for nomenclature of the tourmaline-supergroup minerals, the dominant valence is determined for each site by summing the ions for each valence, and then the dominant cation identified. The dominant ion for each valence determines the species belonging to each garnet group. Detailed nomenclature scheme on determination of the garnet supergroup species was explained in the current IMA-approved report (Grew et al. 2013).

2.2. Data entry of analyses

Upon successful installation of WinGrt, the start-up screen with various pull-down menus and equivalent shortcuts appears. The program allows the user to type garnet analyses (Fig. 1a; wt. %) and selected cation values of biotite, clinopyroxene and orthopyroxene (Fig. 1b; *apfu*) by clicking the *New* icon on the tool bar, selecting the *New File* from the pull-down menu of *File* option or pressing the *Ctrl + N* keys. Entered biotite, clinopyroxene, and orthopyroxene cations (*apfu*) with input pressure (kbar) values are used by program for estimations of garnet–biotite, garnet–clinopyroxene, and garnet–orthopyroxene geothermometers. In the *New File*, *Data Entry Screen*, and *Calculation Screen*, these parameters are highlighted by the ice blue, moon green and faded pink colors, respectively. The standard 30 variables (wt. %) are used by WinGrt for calculation and classification of garnet supergroup mineral analyses in the following order:

Sample No, SiO_2 , TiO_2 , ZrO_2 , HfO_2 , ThO_2 , SnO_2 , Al_2O_3 , Cr_2O_3 , V_2O_5 , Fe_2O_3 , Mn_2O_3 , Sc_2O_3 , $\text{Y}_2\text{O}_3 + \text{REE}_2\text{O}_3$, FeO , MgO , MnO , ZnO , CaO , Na_2O , Li_2O , P_2O_5 , V_2O_5 , Sb_2O_5 , As_2O_5 , Nb_2O_5 , UO_3 , TeO_3 , F and H_2O (wt. %). However, rare earth element oxides (REE_2O_3) analyses from La_2O_3 to Lu_2O_3 (wt. %) can be typed separately in the *Data Entry Screen* section of program.

Garnet analyses entered in an Excel file (“*.xls*” or “*.xlsx*”) in the above order together with biotite, clinopyroxene, orthopyroxene cations (*apfu*) and pressure (kbar) values can be loaded into the program's *Data Entry Screen* by clicking the *Open Excel File* option from the *File* pull-down menu. By selecting the *Edit Excel File* option from the *File* pull-down menu, these can be typed in a blank Excel file (i.e. *MyGarnet*), stored under a different file name, and then loaded into the program's *Data Entry Screen* by clicking the *Open Excel File* option from the *File* pull-down menu. Additional information about data entry or similar topics can be accessed by pressing the F1 function key to display the *WinGrt.chm* file on the screen.

a

Row No	Sample No	SiO ₂	TiO ₂	ZrO ₂	HfO ₂	ThO ₂	SnO ₂	Al ₂ O ₃	Cr ₂ O ₃	V ₂ O ₃	Fe ₂ O ₃	Mn ₂ O ₃	Sc ₂ O ₃	Y ₂ O ₃ +REE ₂ O ₃	FeO	MgO	MnO	ZnO	CaO	Na ₂ O	Li ₂ O	P ₂ O ₅	V ₂ O ₅	Sb ₂ O ₅	As ₂ O ₅	Nb ₂ O ₅	UO ₃	TeO ₃	F	H ₂ O	
1	LT122	36.7						20.9							37.5	4.4	0.24		0.45												
2	PR54	36.5						20.9							37.6	3.8	0.45		0.59												
3	PR40	37.9						19.6							38.3	3.4	0.33		0.78												
4	PR6	37.1						20.6							38.9	3.3	0.28		0.48												
5	P542	37.3						21.3							36.3	4.3	0.55		0.66												
6	PR33	38.3						20.8							34.8	3.1	0.61		3.1												
7	SW163	37.4						21.3							33.1	4	0.92		4.1												
8	SW198	38.4						21.6							33	4	0.47		4												
9	LT115	36.9						20.2							34.4	2.71	0.11		4.6												
10	T172	37.4						21.8							31.3	3.8	0.75		4.7												
11	o1	37.8						20.9							30.5	2.97	0.86		6.4												
12	PF3	37.6						20.8							30.7	3.5	0.32		6.1												
13	SW168	36.9						21							31.2	2.82	0.66		6.9												
14	SW151	36.9						21							29.8	2.93	0.67		8.3												

b

Row No	[P(kbar)Grt-Bt]	[Al(V)]	Ti	Fe ₂₊	Mg]Biotite	[P(kbar)Grt-Cpx]	[Al(total)]	Fe(total)	Fe ₂₊	Mg]Clinopyroxene	[P(kbar)Grt-Opx]	[Al(total)]	Fe ₂₊	Mg]Orthopyroxene
1	5	0.459	0.0455	1.191	1.219	25	0.437	0.191	0.191	0.536	7	0.176	0.891	0.974
2	5	0.4045	0.0955	1.1125	1.3405	25	0.468	0.221	0.221	0.507	7	0.114	0.906	1.008
3	5	0.45	0.082	1.22	1.126	25	0.465	0.228	0.228	0.538	7	0.107	1.173	0.772
4	5	0.402	0.0975	1.1415	1.281	25	0.376	0.216	0.216	0.593	7	0.115	0.858	1.097
5	5	0.3605	0.1105	1.142	1.2955	25	0.388	0.202	0.202	0.664				
6	5	0.3155	0.101	1.1615	1.3075									
7	5	0.4	0.074	0.895	1.54									
8	5	0.3225	0.104	0.7745	1.741									
9	5	0.357	0.0965	1.0545	1.409									
10	5	0.523	0.0605	0.884	1.4435									
11	5	0.3705	0.098	1.105	1.3345									
12	5	0.353	0.083	0.9655	1.501									
13	5	0.537	0.0515	0.733	1.589									
14	5	0.4225	0.093	0.899	1.502									

Fig. 1 Screenshots of the WinGrt *Data Entry Screen* window. **a** – Editing garnet supergroup mineral analyses (wt. %). **b** – Recalculated input biotite, clinopyroxene and orthopyroxene cations (*apfu*) for geothermometer estimations.

3. Worked examples

Using the selected data set from literature, the following examples show how WinGrt can be used in calculation and classification of the garnet supergroup minerals (see Electronic Supplementary Material, ESM 1), as well as in estimation of temperatures by garnet–biotite, garnet–clinopyroxene and garnet–orthopyroxene geothermometers. Once the previously typed or loaded garnet analyses are processed by clicking the *Calculate icon* (i.e. Σ) in the *Data Entry Section* of the program, all output and input parameters are displayed in columns 1–187 (see Tab. 2) of the *Calculation Screen*. Pressing the **Ctrl + F** keys or clicking the *Open File to Calculate* option from the *Calculate* menu also executes the data processing for a selected data file with the extension of “.gsg”. By clicking the *Send results to Excel file* icon in the *Calculation Screen*, all calculations can be stored

in an Excel file (*Output.xlsx*) and then displayed by clicking the *Open and edit Excel file* icon.

Validity of WinGrt outputs has been tested (Tab. 3) with representative garnet supergroup mineral analyses selected from literature (e.g. Munno et al. 1980; Galuskina et al. 2010; Nagashima and Armbruster 2012; Galuskina et al. 2013; Grew et al. 2013; Antao 2014; Ma et al. 2014; Antao and Cruickshank 2018). WinGrt calculates cations (*apfu*) of garnet supergroup mineral analyses (see rows 38–68 in Fig. 2a) and then allocates the recalculated values into the *Z*, *Y*, and *X* sites, respectively (see columns 70–90 in Fig. 2b and 91–111 in Fig. 2c; rows 28–63 in Tab. 3). Anion contents and total of dominant monovalent, divalent, trivalent, tetravalent, pentavalent and hexavalent cations (*apfu*) in the *X*, *Y* and *Z* sites are listed in columns 113–116 and 118–135 of the *Calculation Screen* window (see Fig. 2d). Dominant charge formula and type with garnet group names

Tab. 2 Description of column numbers in the *Calculation Screen* window of WinGrt program and an output Excel file

Row	Explanation	Column numbers
1	Major oxide garnet supergroup mineral analyses (wt. %)	1–31
2	Blank	32
3	Recalculated FeO, Fe ₂ O ₃ , MnO and Mn ₂ O ₃ (wt. %) contents based on the stoichiometric constraints	33–36
4	Blank	37
5	Recalculated cations of garnet supergroup mineral analyses (<i>apfu</i>)	38–68
6	Blank	69
7	Cations allocation at the Z site	70–82
8	Blank	83
9	Cations allocation at the Y site	84–102
10	Blank	103
11	Cations allocation at the X site	104–111
12	Blank	112
13	Anions allocation at the ϕ site	113–116
14	Blank	117
15	Sums of homovalent cations by site and sums of monovalent and divalent anions	118–135
16	Blank	136
17	Dominant charge formula	137
18	Dominant formula type	138
19	Garnet group	139
20	Garnet species	140
21	Blank	141
22	Input <i>P</i> (kbar) and Al ^{VI} , Ti, Fe ²⁺ and Mg (<i>apfu</i>) values of Bt for garnet–biotite geothermometers	142–146
23	Garnet–biotite geothermometers	147–156
24	Blank	157
25	Input <i>P</i> (kbar) and Al _{tot} , Ti, Fe _{tot} and Mg (<i>apfu</i>) values of Cpx for garnet–clinopyroxene geothermometers	158–162
26	Garnet–clinopyroxene geothermometers	163–172
27	Blank	173
28	Input <i>P</i> (kbar) and Al _{tot} , Fe ²⁺ and Mg (<i>apfu</i>) values of Opx for garnet–orthopyroxene geothermometers	174–177
29	Garnet–orthopyroxene geothermometers	178–187

Note: *apfu* = atoms per formula unit; Bt = biotite, Cpx = clinopyroxene, Opx = orthopyroxene

and species according to the nomenclature scheme by Grew et al. (2013) are presented in columns 137–140 of the *Calculation Screen* window (Fig. 3; see rows 68–70 in Tab. 3) for selected mineral analyses from literature. WinGrt calculates Fe²⁺–Mg exchange reaction geothermometers only for a garnet analysis that belongs to the garnet group including almandine, pyrope, spessartine and grossular species. Garnet–biotite geothermometers with input biotite cations (*apfu*) and pressure (kbar) values are presented in columns 142–156 of the *Calculation Screen* window (Fig. 4a). On the other hand, garnet–clinopyroxene and garnet–orthopyroxene geothermometer estimations by WinGrt are listed with input clino- and orthopyroxene cations (*apfu*) and pressure (kbar) values in columns 158–172 (Fig. 4b) and 174–187 (Fig. 4c) of the *Calculation Screen* window, respectively.

3.1. Geothermometry

Estimating the temperature at which minerals crystallized in rocks is an important goal for petrologists and geochemists. Hence, various methods to estimate the *P–T*

conditions of geologic environments have been developed based on laboratory experiments, thermodynamic calculations, and electron-microprobe mineral analyses. Geothermometers that are based on the exchange reactions (e.g. Fe²⁺ and Mg) between different sites in a single mineral or between mineral pairs are widely used in metamorphic rocks involving e.g. olivine, garnet, clinopyroxene, orthopyroxene, spinel, ilmenite, cordierite, biotite, phengite, chlorite and hornblende (Bucher and Grapes 2011). Current version of WinGrt offers the most widely used Fe²⁺–Mg exchange calibrations in garnet–biotite, garnet–clinopyroxene and garnet–orthopyroxene pairs within the application range of greenschist-, amphibolite-, granulite- and eclogite-facies metamorphic rocks.

3.1.1. Garnet–biotite geothermometers

When compared to the other all Fe²⁺–Mg exchange geothermometers, the garnet–biotite is the most popular one due to its applicability to a large variety of rocks and covering a broad range of metamorphic grades from amphibolite- to granulite-facies (see row 21 in Tab. 4). The

Tab. 3 Chemical compositions of selected garnet supergroup minerals recalculated and classified by WinGrT

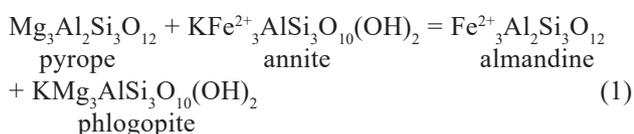
Row		SG1	SG2	SG3	SG4	SG5	SG6	SG7	SG8	SG9	SG10
1	SiO ₂	30.64	31.80	36.60	16.63	25.72	20.90	0.28	0.28	24.64	0.73
2	TiO ₂	1.10	1.76	0.00	6.95	16.00	25.30	2.66	2.62	0.00	0.03
3	ZrO ₂	0.00	0.12	0.00	27.38	2.98	0.00	0.28	4.21	0.00	0.00
4	HfO ₂	0.00	0.00	0.00	0.00	0.00	0.00	0.01	0.00	0.00	0.00
5	SnO ₂	0.00	0.00	0.00	0.00	0.00	0.00	16.65	16.70	0.00	0.00
6	Al ₂ O ₃	4.87	11.20	0.00	9.37	0.72	15.70	11.65	6.17	0.41	0.00
7	Cr ₂ O ₃	0.04	0.00	0.00	0.00	0.01	0.000	0.02	0.00	0.00	0.00
8	V ₂ O ₃	0.17	0.21	0.00	0.00	0.55	0.50	0.00	0.00	0.00	0.00
9	Fe ₂ O ₃	8.38	0.00	0.00	0.00	18.14	0.00	14.38	19.82	0.33	0.00
10	Mn ₂ O ₃	0.00	0.00	0.00	0.00	0.000	0.00	0.00	0.00	31.69	0.00
11	Sc ₂ O ₃	0.17	0.00	0.00	0.00	0.00	0.00	0.00	0.05	0.00	0.00
12	Y ₂ O ₃ +REE ₂ O ₃	25.32	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
13	FeO	11.82	13.50	27.26	6.43	2.21	0.70	1.11	2.20	0.00	0.06
14	MgO	3.98	0.18	0.00	3.32	1.49	2.10	0.22	0.02	0.01	11.69
15	MnO	0.85	0.12	0.00	0.00	0.47	0.000	0.03	0.00	0.00	3.51
16	CaO	13.92	35.80	33.90	30.25	31.63	34.60	25.38	23.86	34.77	20.63
17	Na ₂ O	0.00	0.07	0.00	0.00	0.22	0.00	0.00	0.00	0.00	4.82
18	P ₂ O ₅	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.25
19	V ₂ O ₅	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.33
20	Sb ₂ O ₅	0.00	0.00	0.00	0.00	0.00	0.00	25.98	16.73	0.00	0.00
21	As ₂ O ₅	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.000	0.00	57.40
22	Nb ₂ O ₅	0.00	0.00	0.00	0.00	0.00	0.00	0.07	0.08	0.00	0.00
23	UO ₃	0.00	0.00	0.00	0.00	0.00	0.00	0.05	6.30	0.00	0.00
24	F	0.00	3.13	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
25	O=F	0.00	-1.318	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
26	H ₂ O	0.00	1.93	0.00	0.00	0.00	0.00	0.00	0.00	7.57	0.00
27	Σ (wt. %)	101.26	98.502	97.60	100.33	100.14	99.80	98.79	99.04	99.42	99.45
28	Si	2.762	2.485	3.000	1.507	2.214	1.680	0.031	0.033	1.981	0.070
29	Ti	0.000	0.000	0.000	0.005	0.000	0.000	0.129	0.149	0.000	0.000
30	Al	0.238	0.071	0.000	1.001	0.073	1.320	1.530	0.853	0.000	0.000
31	Fe ³⁺	0.000	0.000	0.000	0.487	0.713	0.000	1.206	1.750	0.000	0.000
32	As	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	2.885
33	V ⁵⁺	0.000	0.000	0.000	0.000	0.000	0.000	0.001	0.000	0.000	0.021
34	P	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.020
35	Fe ²⁺	0.000	0.000	0.000	0.000	0.000	0.000	0.103	0.216	0.000	0.004
36	H ₄	0.000	0.251	0.000	0.000	0.000	0.000	0.000	0.000	1.019	0.000
37	vacancy	0.000	0.193	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
38	Σ Z site	3.000	3.000	3.000	3.000	3.000	3.000	3.000	3.000	3.000	3.000
39	Si	0.000	0.000	0.059	0.000	0.000	0.000	0.000	0.000	0.007	0.000
40	Ti	0.075	0.103	0.000	0.469	1.036	1.530	0.094	0.082	0.000	0.002
41	Zr	0.000	0.005	0.000	1.210	0.125	0.000	0.015	0.241	0.000	0.000
42	Hf	0.000	0.000	0.000	0.000	0.000	0.000	0.0003	0.000	0.000	0.000
43	Sn	0.000	0.000	0.000	0.000	0.000	0.000	0.740	0.781	0.000	0.000
44	Sb	0.000	0.000	0.000	0.000	0.000	0.000	1.075	0.729	0.000	0.000
45	Nb	0.000	0.000	0.000	0.000	0.000	0.000	0.004	0.004	0.000	0.000
46	U	0.000	0.000	0.000	0.000	0.000	0.000	0.001	0.155	0.000	0.000
47	Al	0.279	0.961	0.000	0.000	0.000	0.167	0.000	0.000	0.039	0.000
48	Fe ³⁺	0.568	0.882	1.882	0.000	0.462	0.047	0.000	0.000	0.000	0.000
49	Sc	0.013	0.000	0.000	0.000	0.000	0.000	0.000	0.005	0.000	0.000
50	V ³⁺	0.012	0.013	0.000	0.000	0.038	0.032	0.000	0.000	0.000	0.000
51	Cr	0.003	0.000	0.000	0.000	0.001	0.000	0.002	0.000	0.000	0.000
52	Mn ³⁺	0.000	0.008	0.000	0.000	0.000	0.000	0.000	0.000	1.946	0.000
53	Mg	0.535	0.021	0.000	0.321	0.191	0.244	0.037	0.002	0.001	1.675
54	Fe ²⁺	0.515	0.000	0.023	0.000	0.147	0.000	0.000	0.000	0.000	0.001
55	Mn ²⁺	0.000	0.000	0.000	0.000	0.000	0.000	0.003	0.000	0.000	0.286
56	Σ Y site	2.000	1.993	1.964	2.000	2.000	2.000	1.9703	2.000	1.994	1.964

Tab. 3 Continued

Row	SG1	SG2	SG3	SG4	SG5	SG6	SG7	SG8	SG9	SG10
57 Y+REE	1.214	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
58 Fe ²⁺	0.376	0.000	0.000	0.000	0.012	0.000	0.000	0.000	0.000	0.000
59 Mn ²⁺	0.065	0.000	0.000	0.000	0.034	0.000	0.000	0.000	0.000	0.000
60 Mg	0.000	0.000	0.028	0.128	0.000	0.028	0.000	0.001	0.000	0.000
61 Ca	1.344	2.997	2.980	2.937	2.917	2.980	3.030	2.999	3.006	2.125
62 Na	0.000	0.011	0.000	0.000	0.037	0.000	0.000	0.000	0.000	0.898
63 $\sum X$ site	3.000	3.008	3.008	3.065	3.000	3.008	3.030	3.000	3.006	3.023
64 O	12.000	10.221	12.000	12.000	12.000	12.000	12.000	12.000	7.925	12.000
65 OH	0.000	1.006	0.000	0.000	0.000	0.000	0.000	0.000	4.075	0.000
66 F	0.000	0.773	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
67 \sum Anions	12.000	12.000	12.000	12.000	12.000	12.000	12.000	12.000	12.000	12.000
68 Dominant Formula Type	menzerite	tetravalent Z garnet	schorlomite			bitikleite		henritermierite	berzeliite	
69 Garnet Group	Garnet		Schorlomite			Bitikleite		Henritermierite	Berzeliite	
70 Garnet Species	menzerite-(Y)	grossular	andradite	kimzeyite	schorlomite	hutchonite	bitikleite	dzhuluite	henritermierite	berzeliite

Notes: The formulae were recalculated to 12 anions and 8 cations; Fe³⁺ and Fe²⁺, as well as Mn²⁺ and Mn³⁺, estimations from total FeO (wt. %) contents were carried out on the basis of stoichiometric constraints using Droop's (1987) method; S1, S2, S3 from an Excel spreadsheet developed by Grew et al. (2013) to perform the calculation, cation allocation, species and group determination; S4 from Munno et al. (1980); S5 from Antao (2014); S6 from Ma et al. (2014); S7 from Galuskina et al. (2010); S8 from Galuskina et al. (2013); S9 from Antao and Cruickshank (2018); S10 from Nagashima and Armbruster (2012)

following cation exchange reaction (Eq. 1) has been calibrated by Thompson (1976) as a geothermometer (Eq. 2) based largely on comparison of natural assemblages with experimental phase equilibria:



$$T_{T76}^{\text{Grt-Bt}} (\text{°C}) = \frac{2739.646}{1.56 - \ln K_D} - 273.15 \quad (2)$$

Thompson's (1976) empirical calibration, which is essentially used for natural metapelitic assemblages of low- and medium-grade metamorphic rocks, is based on a linear relationship between $\ln K_D$ and $1/T$, where $K_D = (\text{Mg}^{\text{Grt}} \times \text{Fe}^{\text{Bt}}) / (\text{Fe}^{\text{Grt}} \times \text{Mg}^{\text{Bt}})$. An empirical calibration does not assume any pressure effect on the Fe–Mg fractionation between garnet and biotite phases. However, by clicking the *Use Input P (kbar) for Pressure-independent Garnet–biotite Geothermometer* option from the *Calculate* menu, WinGrt uses previously entered pressure (kbar) values and estimates Thompson's (1976) calibration as a pressure-dependent geothermometer:

$$T_{T76}^{\text{Grt-Bt}} (\text{°C}) = \frac{2740 + 2.34P(\text{kbar})}{1.56 - \ln K_D} - 273.15 \quad (3)$$

Ferry and Spear (1978) presented experimental data at 2.07 kbar and 550–800 °C on the Fe²⁺–Mg exchange reaction between synthetic almandine–pyrope and annite–phlogopite, and proposed an empirical geothermometer (Eq. 4) in systems, where Fe/(Fe + Mg) of garnet was held at 0.9 (see row 22 in Tab. 4):

$$T_{\text{FS78}}^{\text{Grt-Bt}} (\text{°C}) = \frac{2109}{0.782 - \ln K_D} - 273.15 \quad (4)$$

Again, by clicking the *Use Input P (kbar) for Pressure-independent Garnet–Biotite Geothermometer* option from the *Calculate* menu, the program takes into account previously typed pressure (kbar) values and calculates Ferry and Spear's (1978) calibration as pressure-dependent geothermometer (Eq. 5):

$$T_{\text{FS78}}^{\text{Grt-Bt}} (\text{°C}) = \frac{2089 + 9.56P(\text{kbar})}{0.782 - \ln K_D} - 273.15 \quad (5)$$

Hodges and Spear (1982) applied widely used geothermobarometers to pelitic schists of the Mt. Moosilauke region, New Hampshire, where the three aluminum-silicate polymorphs tightly constrain the P – T conditions (see row 23 in Tab. 4). According to Hodges and Spear (1982), previous garnet–biotite calibrations (i.e. Eqs 2 and 4) may be affected by the propensity of reequilibration of garnet and biotite during cooling at geologically reasonable rates, and may underestimate the metamorphic temperatures. Hence, they presented an empirically derived more internally consistent geothermometer for a suite of garnet–biotite–plagioclase–sillimanite–quartz–(\pm andalusite) assemblages from Mt. Moosilauke, New Hampshire:

$$T_{\text{HSS2}}^{\text{Grt-Bt}} (\text{°C}) = \frac{-12454 - 57P(\text{kbar}) - 9900X_{\text{Ca}}^{\text{Grt}}}{3 \times 1.9872 \ln K_D - 4.5X_{\text{Ca}}^{\text{Grt}}} - 273.15 \quad (6)$$

$$\text{where } X_{\text{Ca}}^{\text{Grt}} = \frac{\text{Ca}}{\text{Ca} + \text{Fe}^{2+} + \text{Mg} + \text{Mn}}.$$

In estimating the temperature conditions of biotite–garnet equilibria in metamorphic rocks, Perchuk and Lavrent'eva (1983) proposed a geothermometer using predominantly natural minerals as reactants for their experiments at 550–1000 °C and 5–7 kbar:

WinGrt

Edit Excel Graph About Help

A Windows Program for Garnet Supergroup Minerals *Calculation Screen*

WinGrt	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60	61	62	63	64	65	66	67	68
Sample	[Si]	Ti	Zr	Hf	Th	Sn	Al	Cr	V+3	Fe3+	Mn3+	Sc	Y+REE	Fe2+	Mg	Mn	Zn	Ca	Na	Li	P	V5+	Sb	As	Nb	U	Te	F	O=F	H4	Total(apfu)
Menzerte-(Y)	2.762	0.075	0.000	0.000	0.000	0.000	0.517	0.003	0.012	0.568	0.000	0.013	1.214	0.891	0.535	0.065	0.000	1.344	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	8.000	
Pyrope	2.994	0.049	0.000	0.000	0.000	0.000	1.832	0.011	0.000	0.080	0.000	0.000	0.000	0.537	2.125	0.018	0.000	0.345	0.008	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	8.000	
Grossular	2.931	0.023	0.000	0.000	0.000	0.000	1.937	0.000	0.000	0.089	0.040	0.000	0.000	0.000	0.000	0.005	0.000	2.962	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.013	8.000	
Spessartine	2.876	0.000	0.000	0.000	0.000	0.000	2.063	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	2.831	0.000	0.089	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.117	7.976	
Almandine	2.955	0.075	0.000	0.000	0.000	0.000	2.016	0.000	0.000	0.000	0.000	0.000	0.000	2.519	0.162	0.173	0.000	0.050	0.008	0.000	0.003	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	7.962	
Eringate	2.469	0.453	0.216	0.003	0.000	0.000	0.312	0.070	0.009	0.536	0.000	0.781	0.009	0.035	0.119	0.001	0.000	2.987	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	8.000	
Goldmanite	2.868	0.031	0.000	0.000	0.000	0.000	0.185	0.002	1.821	0.051	0.012	0.000	0.000	0.000	0.000	0.000	0.000	3.096	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	8.066	
Monoite	3.035	0.009	0.000	0.000	0.000	0.000	0.595	0.019	1.331	0.000	0.000	0.000	0.000	0.025	0.010	1.692	0.000	1.267	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	7.983	
Knorringite	3.007	0.006	0.000	0.000	0.000	0.000	0.865	1.041	0.000	0.068	0.000	0.000	0.000	0.411	1.906	0.038	0.000	0.657	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	8.000	
Uvarovite	3.028	0.000	0.000	0.000	0.000	0.000	0.187	1.792	0.000	0.000	0.000	0.000	0.000	0.025	0.061	0.000	0.000	2.887	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	7.982	
Andradite	3.059	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	1.882	0.000	0.000	0.000	0.023	0.000	0.000	0.000	3.036	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	8.000	
Calderite	3.004	0.010	0.000	0.000	0.000	0.000	0.582	0.000	0.000	1.398	0.000	0.000	0.000	0.181	0.006	1.761	0.000	1.055	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	7.997	
Majortite	3.728	0.000	0.000	0.000	0.000	0.000	0.699	0.000	0.000	0.000	0.000	0.000	0.000	0.000	3.496	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	7.922	
Morimotoite	2.319	1.199	0.062	0.000	0.000	0.000	0.098	0.000	0.000	0.740	0.000	0.000	0.000	0.560	0.112	0.017	0.000	2.893	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	8.000	
Kimzeyite	1.507	0.474	1.210	0.000	0.000	0.000	1.001	0.000	0.000	0.487	0.000	0.000	0.000	0.000	0.449	0.000	0.000	2.937	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	8.065	
Innarassite	0.446	0.626	0.410	0.006	0.000	0.000	1.016	1.386	0.000	0.000	0.804	0.000	0.014	0.000	0.070	0.000	0.000	2.965	0.000	0.000	0.000	0.000	0.237	0.000	0.004	0.017	0.000	0.000	0.000	8.000	
Schorlomite	2.214	1.036	0.125	0.000	0.000	0.000	0.073	0.001	0.038	1.175	0.000	0.000	0.000	0.159	0.191	0.034	0.000	2.917	0.037	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	8.000	
Kerimasite	1.656	0.595	0.811	0.017	0.000	0.010	0.465	0.000	0.000	1.303	0.000	0.019	0.002	0.056	0.033	0.000	0.000	3.021	0.000	0.000	0.002	0.000	0.000	0.008	0.000	0.000	0.000	0.000	0.000	8.000	
Toturite	0.552	0.359	0.013	0.000	0.000	0.000	1.462	0.609	0.001	0.000	1.633	0.000	0.000	0.050	0.003	0.000	0.000	2.989	0.000	0.000	0.000	0.001	0.325	0.000	0.002	0.000	0.000	0.000	0.000	8.000	
Hutcheonite	1.680	1.530	0.000	0.000	0.000	0.000	1.487	0.000	0.032	0.047	0.000	0.000	0.000	0.000	0.252	0.000	0.000	2.980	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	8.007	
Bitkietite	0.031	0.223	0.015	0.000	0.000	0.740	1.530	0.002	0.000	1.206	0.000	0.000	0.000	0.103	0.037	0.003	0.000	3.030	0.000	0.000	0.000	0.001	1.075	0.000	0.004	0.001	0.000	0.000	0.000	8.000	
Dzhalulite	0.033	0.231	0.241	0.000	0.000	0.781	0.853	0.000	0.000	1.750	0.000	0.005	0.000	0.216	0.003	0.000	0.000	2.999	0.000	0.000	0.000	0.000	0.729	0.000	0.004	0.155	0.000	0.000	0.000	8.000	
Elbrusite	0.098	0.199	1.040	0.009	0.018	0.255	0.539	0.000	0.000	1.576	0.000	0.000	0.001	0.559	0.004	0.000	0.000	3.040	0.000	0.000	0.000	0.004	0.000	0.000	0.000	0.658	0.000	0.000	0.000	8.000	
Holtzsite	1.862	0.000	0.000	0.000	0.000	0.000	1.293	0.000	0.000	0.000	0.846	0.000	0.000	0.000	0.000	0.000	0.000	3.027	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	1.170	7.998	
Henntermierite	1.988	0.000	0.000	0.000	0.000	0.000	0.039	0.000	0.000	0.000	1.946	0.000	0.000	0.000	0.001	0.000	0.000	3.006	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	1.019	8.000	
Palenzonite	0.183	0.003	0.000	0.000	0.000	0.000	0.002	0.000	0.000	0.000	0.000	0.000	0.000	0.003	0.003	1.818	0.000	2.219	0.771	0.000	0.000	2.712	0.000	0.166	0.000	0.000	0.000	0.000	0.000	7.881	

C:\Program Files (x86)\WinGrt\GarnetGroupMinerals.gsg

Assuming 8 cations Assuming 12 anions 12:15 AM 3/23/2020

WinGrt

Edit Excel Graph About Help

A Windows Program for Garnet Supergroup Minerals *Calculation Screen*

WinGrt	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90
Sample	[Si]	Ti	Al	Fe3+	As	V5+	P	Fe2+	Zn	Li	H4	Vacancy	TotalZ	[Si]	Ti	Zr	Hf	Sn	Sb	Nb	
Menzerte-(Y)	2.762	0.000	0.238	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	3.000	0.000	0.075	0.000	0.000	0.000	0.000	0.000	
Pyrope	2.994	0.000	0.006	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	3.000	0.000	0.049	0.000	0.000	0.000	0.000	0.000	
Grossular	2.931	0.000	0.056	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.013	0.000	3.000	0.000	0.023	0.000	0.000	0.000	0.000	0.000	
Spessartine	2.876	0.000	0.007	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.117	0.000	3.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	
Almandine	2.955	0.000	0.042	0.000	0.000	0.000	0.003	0.000	0.000	0.000	0.000	0.000	3.000	0.000	0.075	0.000	0.000	0.000	0.000	0.000	
Eringate	2.469	0.000	0.312	0.219	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	3.000	0.000	0.453	0.216	0.003	0.000	0.000	0.000	
Goldmanite	2.868	0.000	0.132	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	3.000	0.000	0.031	0.000	0.000	0.000	0.000	0.000	
Monoite	3.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	3.000	0.035	0.009	0.000	0.000	0.000	0.000	0.000	
Knorringite	3.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	3.000	0.007	0.006	0.000	0.000	0.000	0.000	0.000	
Uvarovite	3.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	3.000	0.028	0.000	0.000	0.000	0.000	0.000	0.000	
Andradite	3.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	3.000	0.059	0.000	0.000	0.000	0.000	0.000	0.000	
Calderite	3.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	3.000	0.004	0.010	0.000	0.000	0.000	0.000	0.000	
Majortite	3.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	3.000	0.728	0.000	0.000	0.000	0.000	0.000	0.000	
Morimotoite	2.319	0.000	0.098	0.582	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	3.000	0.000	1.199	0.062	0.000	0.000	0.000	0.000	
Kimzeyite	1.507	0.005	1.001	0.487	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	3.000	0.000	0.469	1.210	0.000	0.00			

Win Grt

Edit Excel Graph About Help

A Windows Program for Garnet Supergroup Minerals Calculation Screen

Win Grt	91	92	93	94	95	96	97	98	99	100	101	102	103	104	105	106	107	108	109	110	111
Sample	Te	U	Al	Fe3+	Sc	V3+	Cr	Mn3+	Mg	Fe2+	Mn2+	TotalY	[Th	Y+REE	Fe2+	Mn2+	Mg	Ca	Na	TotalX	
Menzerite-(Y)	0.000	0.000	0.279	0.568	0.013	0.012	0.003	0.000	0.535	0.515	0.000	2.000		0.000	1.214	0.376	0.065	0.000	1.344	0.000	3.000
Pyrope	0.000	0.000	1.826	0.080	0.000	0.000	0.011	0.000	0.034	0.000	0.000	2.000		0.000	0.000	0.537	0.018	2.091	0.345	0.008	3.000
Grossular	0.000	0.000	1.881	0.089	0.000	0.000	0.000	0.040	0.000	0.000	0.000	2.033		0.000	0.000	0.000	0.005	0.000	2.962	0.000	2.967
Spessartine	0.000	0.000	2.056	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	2.056		0.000	0.000	0.000	2.831	0.000	0.089	0.000	2.920
Almandine	0.000	0.000	1.974	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	2.049		0.000	0.000	2.519	0.173	0.162	0.050	0.008	2.913
Eringate	0.000	0.000	0.000	0.317	0.781	0.009	0.070	0.000	0.119	0.032	0.000	2.000		0.000	0.009	0.003	0.001	0.000	2.987	0.000	3.000
Goldmanite	0.000	0.000	0.053	0.051	0.000	1.821	0.002	0.012	0.000	0.000	0.000	1.970		0.000	0.000	0.000	0.000	0.000	3.096	0.000	3.096
Momolite	0.000	0.000	0.595	0.000	0.000	1.331	0.019	0.000	0.010	0.000	0.000	2.000		0.000	0.000	0.024	1.692	0.000	1.267	0.000	2.983
Knorringite	0.000	0.000	0.865	0.068	0.000	0.000	1.041	0.000	0.013	0.000	0.000	2.000		0.000	0.000	0.411	0.038	1.893	0.657	0.000	3.000
Uvarovite	0.000	0.000	0.187	0.000	0.000	0.000	1.792	0.000	0.000	0.000	0.000	2.008		0.000	0.000	0.025	0.000	0.061	2.887	0.000	2.974
Andradite	0.000	0.000	0.000	1.882	0.000	0.000	0.000	0.000	0.000	0.023	0.000	1.964		0.000	0.000	0.000	0.000	0.000	3.036	0.000	3.036
Calderite	0.000	0.000	0.582	1.398	0.000	0.000	0.000	0.000	0.006	0.000	0.000	2.000		0.000	0.000	0.181	1.761	0.000	1.055	0.000	2.997
Majorite	0.000	0.000	0.699	0.000	0.000	0.000	0.000	0.000	0.573	0.000	0.000	2.000		0.000	0.000	0.000	0.000	2.922	0.000	0.000	2.922
Morimotoite	0.000	0.000	0.000	0.158	0.000	0.000	0.000	0.000	0.112	0.469	0.000	2.000		0.000	0.000	0.091	0.017	0.000	2.893	0.000	3.000
Kimzeyite	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.321	0.000	0.000	2.000		0.000	0.000	0.000	0.000	0.128	2.937	0.000	3.065
Irinarsite	0.000	0.017	0.000	0.000	0.014	0.000	0.000	0.000	0.000	0.000	0.000	2.000		0.000	0.000	0.035	0.000	0.000	2.965	0.000	3.000
Schoriomite	0.000	0.000	0.000	0.462	0.000	0.038	0.001	0.000	0.191	0.147	0.000	2.000		0.000	0.000	0.012	0.034	0.000	2.917	0.037	3.000
Kerimastite	0.000	0.000	0.000	0.426	0.019	0.000	0.000	0.000	0.033	0.056	0.000	1.976		0.000	0.002	0.000	0.000	0.000	3.021	0.000	3.024
Toturite	0.000	0.000	0.000	0.000	0.000	0.000	0.001	0.000	0.000	0.000	0.000	2.000		0.000	0.000	0.007	0.000	0.003	2.989	0.000	3.000
Hutcheonite	0.000	0.000	0.167	0.047	0.000	0.032	0.000	0.000	0.224	0.000	0.000	2.000		0.000	0.000	0.000	0.000	0.028	2.980	0.000	3.007
Bitkileite	0.000	0.001	0.000	0.000	0.000	0.002	0.000	0.037	0.000	0.003	1.970			0.000	0.000	0.000	0.000	0.000	3.030	0.000	3.030
Dzhuluite	0.000	0.155	0.000	0.000	0.005	0.000	0.000	0.000	0.002	0.000	0.000	2.000		0.000	0.000	0.000	0.000	0.001	2.999	0.000	3.000
Elbrusite	0.000	0.658	0.000	0.000	0.000	0.000	0.000	0.000	0.004	0.000	0.000	1.966		0.018	0.001	0.000	0.000	0.000	3.040	0.000	3.059
Holtstamite	0.000	0.000	1.293	0.000	0.000	0.000	0.000	0.646	0.000	0.000	0.000	1.971		0.000	0.000	0.000	0.000	0.000	3.027	0.000	3.027
Henritermierite	0.000	0.000	0.039	0.000	0.000	0.000	0.000	1.946	0.001	0.000	0.000	1.994		0.000	0.000	0.000	0.000	0.000	3.006	0.000	3.006
Palenzonite	0.000	0.000	0.002	0.000	0.000	0.000	0.000	0.000	0.003	0.003	1.818	1.891		0.000	0.000	0.000	0.000	0.000	2.219	0.771	2.990

C:\Program Files (x86)\WinGrt\GarnetGroupMinerals.gsg

Assuming 8 cations Assuming 12 anions 12:15 AM 3/23/2020

Win Grt

Edit Excel Graph About Help

A Windows Program for Garnet Supergroup Minerals Calculation Screen

Win Grt	118	119	120	121	122	123	124	125	126	127	128	129	130	131	132	133
Sample	Monovalent Z	Divalent Z	Trivalent Z	Tetravalent Z	Pentavalent Z	Hydrogen for Z	Vacancy at Z	Divalent Y	Trivalent Y	Tetravalent Y	Pentavalent Y	Hexavalent Y	Monovalent X	Divalent X	Trivalent X	Tetravalent X
Menzerite-(Y)	0.00	0.00	0.24	2.76	0.00	0.00	0.00	1.05	0.88	0.07	0.00	0.00	0.00	1.79	1.21	0.00
Pyrope	0.00	0.00	0.01	2.99	0.00	0.00	0.00	0.03	1.92	0.05	0.00	0.00	0.01	2.99	0.00	0.00
Grossular	0.00	0.00	0.06	2.93	0.00	0.01	0.00	0.00	2.01	0.02	0.00	0.00	0.00	2.97	0.00	0.00
Spessartine	0.00	0.00	0.01	2.88	0.00	0.12	0.00	0.00	2.06	0.00	0.00	0.00	0.00	2.92	0.00	0.00
Almandine	0.00	0.00	0.04	2.96	0.00	0.00	0.00	0.00	1.97	0.07	0.00	0.00	0.01	2.91	0.00	0.00
Eringate	0.00	0.00	0.53	2.47	0.00	0.00	0.00	0.15	1.18	0.67	0.00	0.00	0.00	2.99	0.01	0.00
Goldmanite	0.00	0.00	0.13	2.87	0.00	0.00	0.00	0.00	1.94	0.03	0.00	0.00	0.00	3.10	0.00	0.00
Momolite	0.00	0.00	0.00	3.00	0.00	0.00	0.00	0.01	1.95	0.04	0.00	0.00	0.00	2.98	0.00	0.00
Knorringite	0.00	0.00	0.00	3.00	0.00	0.00	0.00	0.01	1.97	0.01	0.00	0.00	0.00	3.00	0.00	0.00
Uvarovite	0.00	0.00	0.00	3.00	0.00	0.00	0.00	0.00	1.98	0.03	0.00	0.00	0.00	2.97	0.00	0.00
Andradite	0.00	0.00	0.00	3.00	0.00	0.00	0.00	0.02	1.88	0.06	0.00	0.00	0.00	3.04	0.00	0.00
Calderite	0.00	0.00	0.00	3.00	0.00	0.00	0.00	0.01	1.98	0.01	0.00	0.00	0.00	3.00	0.00	0.00
Majorite	0.00	0.00	0.00	3.00	0.00	0.00	0.00	0.57	0.70	0.73	0.00	0.00	0.00	2.92	0.00	0.00
Morimotoite	0.00	0.00	0.68	2.32	0.00	0.00	0.00	0.58	0.16	1.26	0.00	0.00	0.00	3.00	0.00	0.00
Kimzeyite	0.00	0.00	1.49	1.51	0.00	0.00	0.00	0.32	0.00	1.68	0.00	0.00	0.00	3.06	0.00	0.00
Irinarsite	0.00	0.04	2.19	0.77	0.00	0.00	0.00	0.00	0.01	1.73	0.24	0.02	0.00	3.00	0.00	0.00
Schoriomite	0.00	0.00	0.79	2.21	0.00	0.00	0.00	0.34	0.50	1.16	0.00	0.00	0.04	2.96	0.00	0.00
Kerimastite	0.00	0.00	1.34	1.66	0.00	0.00	0.00	0.09	0.45	1.43	0.01	0.00	0.00	3.02	0.00	0.00
Toturite	0.00	0.04	2.24	0.71	0.00	0.00	0.00	0.00	0.00	1.67	0.33	0.00	0.00	3.00	0.00	0.00
Hutcheonite	0.00	0.00	1.32	1.68	0.00	0.00	0.00	0.22	0.25	1.53	0.00	0.00	0.00	3.01	0.00	0.00
Bitkileite	0.00	0.10	2.74	0.16	0.00	0.00	0.00	0.04	0.00	0.85	1.08	0.00	0.00	3.03	0.00	0.00
Dzhuluite	0.00	0.22	2.60	0.18	0.00	0.00	0.00	0.00	0.01	1.10	0.73	0.16	0.00	3.00	0.00	0.00
Elbrusite	0.00	0.56	2.11	0.30	0.00	0.00	0.00	0.00	0.00	1.30	0.00	0.66	0.00	3.04	0.00	0.02
Holtstamite	0.00	0.00	0.00	1.83	0.00	1.17	0.00	0.00	1.94	0.03	0.00	0.00	0.00	3.03	0.00	0.00
Henritermierite	0.00	0.00	0.00	1.98	0.00	1.02	0.00	0.00	1.99	0.01	0.00	0.00	0.00	3.01	0.00	0.00
Palenzonite	0.00	0.00	0.00	0.12	2.88	0.00	0.00	1.82	0.00	0.06	0.00	0.00	0.77	2.22	0.00	0.00

C:\Program Files (x86)\WinGrt\GarnetGroupMinerals.gsg

Assuming 8 cations Assuming 12 anions 12:15 AM 3/23/2020

Fig. 2 Continued. c – Cations allocation at the Z, Y, and X sites (apfu). d – Sums of homovalent cations by site (apfu).

Win Grt	137	138	139	140	141	142	143	144	145	146	147	148	149
Sample	[Dominant Charge Formula]	[Dominant Formula Type]	[Garnet Group]	[Garnet Species]	[P](kbar)	[AVI]	Ti	Fe2+	Mg(apfu)	[TT76]	[TFS78]	[THSI]	
Menzerite-(Y)	{X3+2X2+}[Y2+2](Z4+3)A2-12	menzerite	Garnet	menzerite-(Y)									
Pyrope	{X2+3}[Y3+2](Z4+3)A2-12	tetravalent Z garnet	Garnet	pyrope									
Grossular	{X2+3}[Y3+2](Z4+3)A2-12	tetravalent Z garnet	Garnet	grossular									
Spessartine	{X2+3}[Y3+2](Z4+3)A2-12	tetravalent Z garnet	Garnet	spessartine									
Almandine	{X2+3}[Y3+2](Z4+3)A2-12	tetravalent Z garnet	Garnet	almandine									
Eringite	{X2+3}[Y3+2](Z4+3)A2-12	tetravalent Z garnet	Garnet	eringite									
Goldmanite	{X2+3}[Y3+2](Z4+3)A2-12	tetravalent Z garnet	Garnet	goldmanite									
Moroite	{X2+3}[Y3+2](Z4+3)A2-12	tetravalent Z garnet	Garnet	moroite									
Knorringite	{X2+3}[Y3+2](Z4+3)A2-12	tetravalent Z garnet	Garnet	knorringite									
Uvarovite	{X2+3}[Y3+2](Z4+3)A2-12	tetravalent Z garnet	Garnet	uvarovite									
Andradite	{X2+3}[Y3+2](Z4+3)A2-12	tetravalent Z garnet	Garnet	andradite									
Caldierite	{X2+3}[Y3+2](Z4+3)A2-12	tetravalent Z garnet	Garnet	caldierite									
Majortite	{X2+3}[Y4+Y2+](Z4+3)A2-12	morimotoite	Garnet	majortite									
Morimotoite	{X2+3}[Y4+Y2+](Z4+3)A2-12	morimotoite	Garnet	morimotoite									
Kimzeyite	{X2+3}[Y4+2](Z4+Z3+2)A2-12	schorlomite	Schorlomite	kimzeyite									
Irinassite	{X2+3}[Y4+2](Z4+Z3+2)A2-12	schorlomite	Schorlomite	irinassite									
Schorlomite	{X2+3}[Y4+2](Z4+Z3+2)A2-12	schorlomite	Schorlomite	schorlomite									
Kerimasite	{X2+3}[Y4+2](Z4+Z3+2)A2-12	schorlomite	Schorlomite	kerimasite									
Toturite	{X2+3}[Y4+2](Z4+Z3+2)A2-12	schorlomite	Schorlomite	toturite									
Hutcheonite	{X2+3}[Y4+2](Z4+Z3+2)A2-12	schorlomite	Schorlomite	hutcheonite									
Bitkileite	{X2+3}[Y5+Y4+](Z3+3)A2-12	bitkileite	Bitkileite	bitkileite									
Dzhululite	{X2+3}[Y5+Y4+](Z3+3)A2-12	bitkileite	Bitkileite	dzhululite									
Elbrusite	{X2+3}[Y4+1.5Y6+0.5](Z3+3)A2-12	elbrusite	Bitkileite	elbrusite									
Holstamite	{X2+3}[Y3+2](Z4+2)A2-8(OH)4	henritermieite	Henritermieite	holstamite									
Henritermieite	{X2+3}[Y3+2](Z4+2)A2-8(OH)4	henritermieite	Henritermieite	henritermieite									
Palenzonaite	{X2+2X1+}[Y2+2](Z5+3)A2-12	berzelite	Berzelite	palenzonaite									

Fig. 3 Screenshots of the WinGrt Calculation Screen window showing the dominant charge formula, formula type, garnet group and garnet species.

$$T_{PL83}^{Grt-Bt} (^{\circ}C) = \frac{7843.7 - 0.0246P(\text{kbar}) - 6}{1.987 \ln K_D + 5.699} - 273.15 \quad (7)$$

Taking into account the negative Clausius–Clapeyron slopes, Thoenen (1989) suggested a corrected version of garnet–biotite geothermometer developed by Perchuk and Lavrent'eva (1983):

$$T_{PL83}^{Grt-Bt} (^{\circ}C) = \frac{3890 + 9.56P(\text{kbar})}{2.868 - \ln K_D} - 273.15 \quad (8)$$

WinGrt uses this revised version of equation by Thoenen (1989) in estimation of the Perchuk and Lavrent'eva (1983) calibration (see row 24 in Tab. 4).

Considering a reliable temperature estimate for a wide compositional range of natural garnet and biotite assemblages, Ganguly and Saxena (1984) extensively studied the mixing behavior of non-ideality in quaternary (Fe–Mg–Ca–Mn) garnet solid solution and an ideal mixing of Fe–Mg in biotite. Thus, on the basis of thermodynamic and statistical treatment of natural data, they formulated the following garnet–biotite geothermometer (see row 25 in Tab. 4):

$$T_{GS84}^{Grt-Bt} (^{\circ}C) = \frac{-17371 - 79.5P(\text{kbar}) + 0.8 \times 9497.7}{8.3144 \ln K_D - 0.782} - \frac{-W_{MgFe}^{Grt} (X_{Fe}^{Grt} - X_{Mg}^{Grt}) - 12552 X_{Ca}^{Grt} - 12552 X_{Mn}^{Grt}}{8.3144 \ln K_D - 0.782} - 273.15 \quad (9)$$

$$\text{where } W_{MgFe}^{Grt} = \frac{837 X_{Mg}^{Grt} + 10460 X_{Fe}^{Grt}}{X_{Mg}^{Grt} + X_{Fe}^{Grt}} \text{ and } X_i^{Grt} = \frac{i}{Mg + Fe^{2+} + Mn + Ca}.$$

Since the first attempts of empirical calibration of distribution coefficient (K_D) as a function of temperature, biotite–garnet geothermometers (e.g. Thompson 1976; Ferry and Spear 1978) have been widely used in estimation of temperature conditions in medium- to high-grade pelitic and semipelitic rocks. However, application of these empirical calibrations to high-grade metamorphic terranes, such as granulite-facies rocks, showed internally inconsistent and highly variable results (Bohlen and Essene 1980). Indares and Martignole (1985) reevaluated the various garnet–biotite geothermometers in granulite-facies rocks and, based on their own observations and analyses, provided two improved calibrations (Eqs 10 and 11) of Ferry and Spear's (1978) garnet–biotite geothermometer (see rows 26–27 in Tab. 4) that take into account the effects of Ti and Al in the biotite solid-solution:

$$T_{IM85}^{Grt-Bt} (^{\circ}C) = \frac{12454 + 57P(\text{kbar}) - 3 \times 1590(AI_{Bt}^{VI}/3)}{4.662 - 5.9616 \ln K_D} + \frac{3 \times 7451(Ti_{Bt}/3) + 9000(X_{Ca}^{Grt} + X_{Mn}^{Grt})}{4.662 - 5.9616 \ln K_D} - 273.15 \quad (10)$$

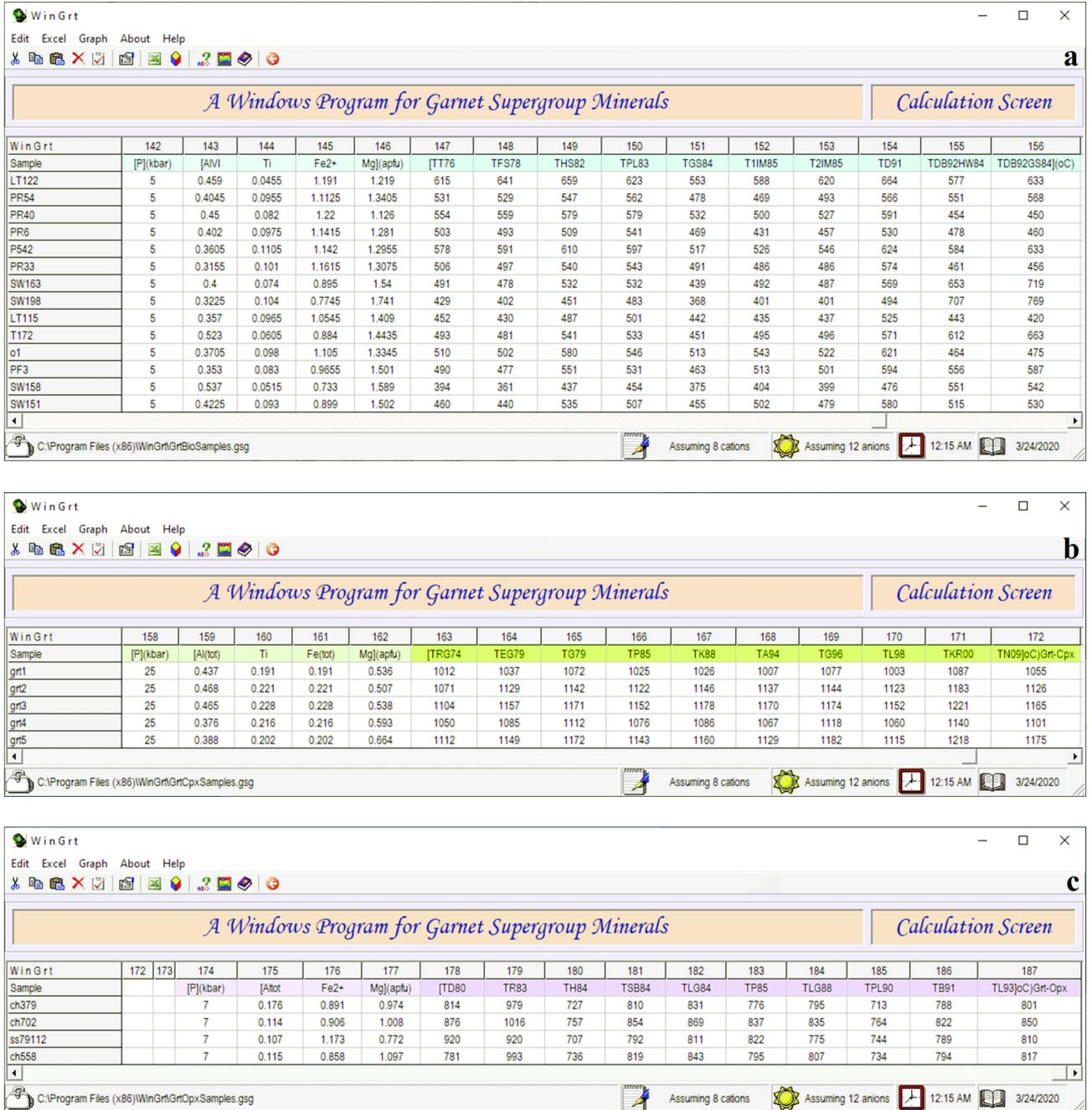


Fig. 4 Screenshots of the WinGrt Calculation Screen window. **a** – Input P (kbar) and Al^{VI} , Ti, Fe^{2+} , and Mg (apfu) values of biotite for garnet-biotite geothermometers. **b** – Input P (kbar) and Al^{tot} , Ti, Fe_{tot} and Mg (apfu) values of clinopyroxene for garnet-clinopyroxene geothermometers. **c** – Input P (kbar) and Al^{tot} , Fe^{2+} and Mg (apfu) values of orthopyroxene for garnet-orthopyroxene geothermometers.

$$T_{2IM85}^{Grt-Bt} (°C) = \frac{12454 + 57P(kbar) + 3 \times 454(Al_{Bt}^{VI}/3)}{4.662 - 5.9616 \ln K_D + 4.5X_{Ca}^{Grt}} + \frac{3 \times 6767(Ti_{Bt}/3) + 9000X_{Ca}^{Grt}}{4.662 - 5.9616 \ln K_D + 4.5X_{Ca}^{Grt}} - 273.15 \quad (11)$$

Dasgupta et al. (1991) derived a new formulation of garnet-biotite Fe^{2+} -Mg exchange geothermometer through statistical regression of the reversed experimental data

of Ferry and Spear (1978) taking into consideration the non-ideality terms in garnet and biotite solid solutions. According to Dasgupta et al. (1991), the resultant geothermometer (Eq. 12; see row 28 in Tab. 4) gives consistent results for rocks with a much wider compositional range when compared to the earlier garnet-biotite formulations:

$$T_{D91}^{Grt-Bt} (°C) = \frac{4301 + 3000X_{Ca}^{Grt} + 1300X_{Mn}^{Grt}}{1.85 - 1.9872 \ln K_D}$$

Tab. 4 Garnet–biotite geothermometer estimations by WinGrt program

Row		SG1	SG2	SG3	SG4	SG5	SG6	SG7	SG8	SG9	SG10	SG11
1	SiO ₂	36.700	36.500	37.900	37.100	37.300	38.300	37.400	38.400	36.900	37.400	37.800
2	Al ₂ O ₃	20.900	20.900	19.600	20.600	21.300	20.800	21.300	21.600	20.200	21.800	20.900
3	FeO	37.500	37.600	38.300	38.900	36.300	34.800	33.100	33.000	34.400	31.300	30.500
4	MgO	4.400	3.800	3.400	3.300	4.300	3.100	4.000	4.000	2.710	3.800	2.970
5	MnO	0.240	0.450	0.330	0.280	0.550	0.610	0.920	0.470	0.110	0.750	0.860
6	CaO	0.450	0.590	0.780	0.480	0.660	3.100	4.100	4.000	4.600	4.700	6.400
7	Σ (wt. %)	100.190	99.840	100.310	100.660	100.410	100.710	100.820	101.470	98.920	99.750	99.430
Garnet recalculation based on 8 cations (apfu)												
8	Si	2.938	2.942	3.056	2.980	2.974	3.045	2.953	3.007	2.996	2.974	3.023
9	Al	1.972	1.985	1.863	1.950	2.001	1.949	1.982	1.994	1.933	2.043	1.970
10	Fe ³⁺	0.153	0.131	0.025	0.089	0.051	0.000	0.112	0.000	0.076	0.009	0.000
11	Fe ²⁺	2.358	2.404	2.557	2.525	2.369	2.314	2.074	2.161	2.260	2.073	2.040
12	Mg	0.525	0.457	0.409	0.395	0.511	0.367	0.471	0.467	0.328	0.450	0.354
13	Mn	0.016	0.031	0.023	0.019	0.037	0.041	0.062	0.031	0.008	0.051	0.058
14	Ca	0.039	0.051	0.067	0.041	0.056	0.264	0.347	0.336	0.400	0.400	0.548
15	Σ cations	8.000	8.000	8.000	8.000	8.000	7.980	8.000	7.996	8.000	8.000	7.993
Garnet species												
	Alm	Alm	Alm	Alm	Alm	Alm	Alm	Alm	Alm	Alm	Alm	Alm
Input recalculated biotite cations (apfu) and pressure (kbar) values for garnet–biotite geothermometers												
16	Al ^{VI}	0.459	0.405	0.450	0.402	0.361	0.316	0.400	0.323	0.357	0.523	0.371
17	Ti	0.046	0.096	0.082	0.098	0.111	0.101	0.074	0.104	0.097	0.061	0.098
18	Fe ²⁺	1.191	1.113	1.220	1.142	1.142	1.162	0.895	0.775	1.055	0.884	1.105
19	Mg	1.219	1.341	1.126	1.281	1.296	1.308	1.540	1.741	1.409	1.444	1.335
20	P (kbar)	5	5	5	5	5	5	5	5	5	5	5
Garnet–biotite geothermometers (T, °C)												
21	T1	619	535	557	507	582	509	494	432	455	496	513
22	T2	653	540	570	504	602	507	488	411	439	491	512
23	T3	659	547	579	509	610	540	532	451	487	541	580
24	T4	623	562	579	541	597	543	532	483	501	533	546
25	T5	553	478	532	469	517	491	439	368	442	451	513
26	T6	588	469	500	431	526	486	492	401	435	495	543
27	T7	620	493	527	457	546	486	487	401	437	496	522
28	T8	664	566	591	530	624	574	569	494	525	571	621
29	T9	577	551	454	478	584	461	653	707	443	612	464
30	T10	633	568	450	460	633	456	719	769	420	663	475

Notes: The formulae were recalculated to 12 anions and 8 cations; Fe³⁺ and Fe²⁺ estimations from total FeO (wt. %) contents were carried out on the basis of stoichiometric constraints using Droop's (1987) method; sample numbers SG1 to SG11 from Hoinkes (1986);

Alm = almandine; Thermometers T1 by Thompson (1976), T2 by Ferry and Spear (1978), T3 by Hodges and Spear (1982), T4 by Perchuk and Lavrent'eva (1983), T5 by Ganguly and Saxena (1984), T6 and T7 by Indares and Mortignale (1985), T8 by Dasgupta et al. (1991), T9 by Bhattacharya et al. (1992) using the updated mixing parameters of Hackler and Wood (1989), T10 by Bhattacharya et al. (1992) using the updated mixing parameters of Ganguly and Saxena (1984)

$$\frac{-495(X_{\text{Mg}}^{\text{Grt}} - X_{\text{Fe}}^{\text{Grt}}) - 3595(\text{Al}_{\text{Bt}}^{\text{VI}}/3) - 4423(\text{Ti}_{\text{Bt}}/3)}{1.85 - 1.9872 \ln K_D} + \frac{+1073((\text{Mg}_{\text{Bt}}/3) - (\text{Fe}_{\text{Bt}}/3)) + 24.6P(\text{kbar})}{1.85 - 1.9872 \ln K_D} - 273.15 \quad (12)$$

Bhattacharya et al. (1992) attempted a theoretical analysis of the experimentally observed P – T – X (Mg–Fe) relationship in the biotite–garnet system using the updated mixing parameters for the pyrope–almandine asymmetric regular solution by Ganguly and Saxena (1984) and Hackler and Wood (1989). Thus, they proposed two new garnet–biotite formulations (see rows 29–30 in

Tab. 4) for application to natural rocks including garnet and biotite pairs:

$$T1_{\text{B92_GS84}}^{\text{Grt-Bt}} (\text{°C}) = \frac{13538 + 19.3P(\text{kbar}) - \{837(X_{\text{Mg}}^{\text{Grt}})^2\}}{6.778 + 8.3143 \ln K_D + 6.276X_{\text{Ca}}^{\text{Grt}}(1 - X_{\text{Mn}}^{\text{Grt}})} - \frac{10460(X_{\text{Fe}}^{\text{Grt}})^2 - 13807X_{\text{Ca}}^{\text{Grt}}(1 - X_{\text{Mn}}^{\text{Grt}})}{6.778 + 8.3143 \ln K_D + 6.276X_{\text{Ca}}^{\text{Grt}}(1 - X_{\text{Mn}}^{\text{Grt}})} + \frac{19246X_{\text{Fe}}^{\text{Grt}}X_{\text{Mg}}^{\text{Grt}}(1 - X_{\text{Mn}}^{\text{Grt}}) + 5649X_{\text{Ca}}^{\text{Grt}}(X_{\text{Mg}}^{\text{Grt}} - X_{\text{Fe}}^{\text{Grt}})}{6.778 + 8.3143 \ln K_D + 6.276X_{\text{Ca}}^{\text{Grt}}(1 - X_{\text{Mn}}^{\text{Grt}})} + \frac{7972(2X_{\text{Mg}}^{\text{Bt}} - 1)}{6.778 + 8.3143 \ln K_D + 6.276X_{\text{Ca}}^{\text{Grt}}(1 - X_{\text{Mn}}^{\text{Grt}})} - 273.15 \quad (13)$$

$$T_{B92_HW84}^{Grt-Bt}(\text{°C}) = \frac{20286 + 19.3P(\text{kbar}) - \{2080(X_{Mg}^{Grt})^2 - 6350(X_{Fe}^{Grt})^2 - 13807X_{Ca}^{Grt}(1 - X_{Mn}^{Grt})\}}{13.138 + 8.3143 \ln K_D + 6.276X_{Ca}^{Grt}(1 - X_{Mn}^{Grt})} - 273.15 \quad (14)$$

where $K_D = \text{FeMg}_{Grt}/\text{FeMg}_{Bt}$; $\text{FeMg}_{Grt} = \text{Fe}_{Grt}/\text{Mg}_{Grt}$; $\text{FeMg}_{Bt} = \text{Mg}_{Bt}/\text{Fe}_{Bt}$; $\text{Fe}_{Grt} = \text{Fe}^{2+}/(\text{Mg} + \text{Fe}^{2+} + \text{Mn} + \text{Ca})$; $\text{Mg}_{Grt} = \text{Mg}/(\text{Mg} + \text{Fe}^{2+} + \text{Mn} + \text{Ca})$; Mg_{Bt} = octahedral Mg content (*apfu*) and Fe_{Bt} = octahedral Fe^{2+} content (*apfu*).

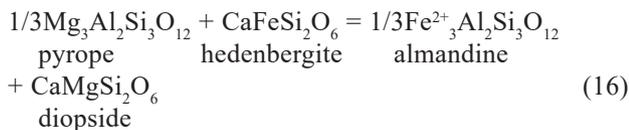
Consequently, the garnet–biotite geothermometer appears to estimate the temperature conditions of rocks metamorphosed under high-*T* greenschist and amphibolite facies fairly well. However, due to the retrograde Fe^{2+} –Mg exchange in the upper amphibolite and granulite facies, the garnet–biotite geothermometers may give anomalously low temperatures, if garnet rims and adjacent biotites in thin sections are analyzed. In this case, reasonable estimates of maximum prograde temperatures may be obtained by using the garnet core and matrix biotite compositions, provided garnet and biotite compositions were once in chemical equilibrium (Bucher and Grapes 2011).

3.1.2. Garnet–clinopyroxene geothermometers

Garnet–clinopyroxene geothermometers which were in general calibrated in the range of 600–1500 °C, have been widely used by petrologists and geochemists. Without consideration of the non-ideality of garnet, Råheim and Green (1974) experimentally calibrated K_D as a function of pressure and temperature for basaltic rocks that crystallized to eclogite at 20–40 kbar and 600–1400 °C, respectively. Based on the experimental results, they proposed an empirical garnet–clinopyroxene geothermometer (see row 25 in Tab. 5) that could only be used for rocks of basaltic composition:

$$T_{RG74}^{Grt-Cpx}(\text{°C}) = \frac{3686 + 28.35P(\text{kbar})}{2.33 + \ln K_D} - 273.15 \quad (15)$$

where $K_D = (\text{Fe}^{2+}/\text{Mg})_{Grt}/(\text{Fe}^{2+}/\text{Mg})_{Cpx}$ for the following garnet–clinopyroxene Fe^{2+} –Mg exchange reaction:



Ellis and Green (1979) carried out a series of experiments for basaltic compositions within the simple CaO – MgO – FeO – Al_2O_3 – SiO_2 system that crystallized to garnet–clinopyroxene-bearing mineral assemblages at 24–30 kbar and 750–1300 °C. They experimentally proven that the non-ideality of Ca content in garnet is important, and thus incorporated a non-ideal term for Ca in garnet into their formulation. Ellis and Green (1979) established K_D as a function of pressure, temperature, and molar fraction of grossular (X_{Ca}^{Grt}), proposing an empirical geothermometer (see row 26 in Tab. 5) applicable to garnet–clinopyroxene-bearing rocks from a wide range of geological environments:

$$T_{EG79}^{Grt-Cpx}(\text{°C}) = \frac{3104X_{Ca}^{Grt} + 3030 + 10.86P(\text{kbar})}{\ln K_D + 1.9034} - 273.15 \quad (17)$$

Being an improved version of the first garnet–clinopyroxene geothermometer (i.e. Råheim and Green 1974), Eq. 17 and its different calibrations found a broad application in the field of metamorphic petrology.

Ganguly (1979) developed a semiempirical theoretical formulation of garnet–clinopyroxene geothermometer (see row 27 in Tab. 5) on the basis of thermodynamic mixing data of garnet solid solutions with grossular and spessartine components:

$$T_{G79}^{Grt-Cpx}(\text{°C}) = \frac{4100 + 1586X_{Ca}^{Grt} + 11.07P(\text{kbar})}{2.4 + \ln K_D} - 273.15 \quad (18a)$$

for $T(\text{°C}) \geq 1060$

$$T_{G79}^{Grt-Cpx}(\text{°C}) = \frac{4801 + 1586X_{Ca}^{Grt} + 11.07P(\text{kbar})}{2.93 + \ln K_D} - 273.15 \quad (18b)$$

for $T(\text{°C}) \leq 1060$

WinGrt estimates temperatures by Ganguly's (1979) geothermometer using the Eq. 18a as a default. However, by clicking the *Use Ganguly's (1979) Garnet–Clinopyroxene Geothermometer for $T(\text{°C}) \leq 1060$* option from the *Calculate* menu, the program calculates Ganguly's (1979) geothermometer using the Eq. 18b.

The process of calibration of a geothermobarometer should involve the determination of the parameters, as well as the uncertainties. Powell (1985) proposed a new garnet–clinopyroxene geothermometer (see row 28 in Tab. 5), derived using a rigorous statistical consideration:

$$T_{P85}^{Grt-Cpx}(\text{°C}) = \frac{2790 + 3140X_{Ca}^{Grt} + 10P(\text{kbar})}{1.735 + \ln K_D} - 273.15 \quad (19)$$

Using the previous experimental data on the partitioning of Fe^{2+} and Mg between garnet and clinopyroxene (e.g. Råheim and Green 1974; Ellis and Green 1979), a new expression for the garnet–clinopyroxene geother-

momenter (Eq. 20; see row 29 in Tab. 5), including a curvilinear relationship between $\ln K_D$ and molar fraction of grossular (X_{Ca}^{Grt}), has been derived by Krogh (1988). It was used to calculate equilibration temperatures for samples of eclogites and associated omphacite-bearing high-pressure gneisses from the Tromsø nappe complex within the northern Scandinavian Caledonides.

$$T_{K88}^{Grt-Cpx} (^{\circ}C) = \frac{-6173(X_{Ca}^{Grt})^2 + 6731X_{Ca}^{Grt} + 1879 + 10P(\text{kbar})}{1.393 + \ln K_D} - 273.15 \quad (20)$$

Taking into account the significant influence of garnet Mg number (Mg#) on the garnet–clinopyroxene Fe²⁺/Mg partition coefficient (K_D), Ai (1994) used a comprehensive experimental dataset and developed an empirical geothermometer (see row 30 in Tab. 5) for the lower crustal rocks including garnet amphibolites, granulites and eclogites as well as for the upper mantle eclogite and lherzolite xenoliths in kimberlites and mineral inclusions in diamonds:

$$T_{K88}^{Grt-Cpx} (^{\circ}C) = \frac{-1629(X_{Ca}^{Grt})^2 + 3648.55X_{Ca}^{Grt} - 6.59Mg\#^{Grt}}{1.076 + \ln K_D} + \frac{+1987.98 + 17.66P(\text{kbar})}{1.076 + \ln K_D} - 273.15 \quad (21)$$

where $Mg\#^{Grt} = 100Mg/(Mg + Fe^{2+})$.

Ganguly et al. (1996) reported new experimental data involving primarily Mg–Ca–Mn and Fe–Mg–Ca–Mn and to a limited extent Fe–Mg–Ca and Mg–Ca garnet compositions. They showed phase diagrams to illustrate unmixing in the binary and ternary garnet solid solutions, presented thermodynamically based formulations, and applied the solution model to develop compositional corrections for the garnet–orthopyroxene and garnet–clinopyroxene Fe²⁺–Mg exchange geothermometers (see row 31 in Tab. 5):

$$T_{G96}^{Grt-Cpx} (^{\circ}C) = \frac{4100 + 11.07P(\text{kbar}) - A}{2.4 + \ln K_D + B} - 273.15 \quad (22)$$

where

$$A = \frac{(21627 + 12P)(-2X_{Ca}^{Grt}X_{Mg}^{Grt} + a)}{8.3144} + \frac{(9834 + 58P)((-X_{Ca}^{Grt})^2 + a) + 873(2X_{Ca}^{Grt}X_{Fe}^{Grt} + a)}{8.3144} - \frac{(6733 + 30P)((-X_{Ca}^{Grt})^2 + b) + (12083 - 539)X_{Mn}^{Grt}}{8.3144},$$

$$B = \frac{-5.78(-2X_{Ca}^{Grt}X_{Mg}^{Grt} + a) - 5.78(-X_{Ca}^{Grt})^2 + a}{8.3144} - \frac{1.69(2X_{Ca}^{Grt}X_{Fe}^{Grt} + 100) + 1.69((-X_{Ca}^{Grt})^2 + b) + 7.69X_{Mn}^{Grt}}{8.3144},$$

$$a = 0.5X_{Ca}^{Grt}(X_{Mg}^{Grt} - X_{Fe}^{Grt} - X_{Mn}^{Grt}) \text{ and } b = 0.5X_{Ca}^{Grt}(X_{Mg}^{Grt} - X_{Fe}^{Grt} + X_{Mn}^{Grt}).$$

They also argued that their garnet–clinopyroxene geothermometer should not be applied to eclogites unless these are mantle-derived (i.e. $T > 1000^{\circ}C$).

Liu (1998) carried out a series of experimental studies for Fe–Mg exchange equilibrium between garnet and clinopyroxene at 600–950 °C, 0.8–3.0 GPa, and fayalite–quartz–magnetite buffer (fO_2) conditions in the basalt–H₂O system. He formulated a new garnet–clinopyroxene geothermometer (see row 32 in Tab. 5) applicable to rocks in amphibolite, granulite, and eclogite terranes:

$$T_{L98}^{Grt-Cpx} (^{\circ}C) = \frac{3820}{1 + 0.13 \times (2.2 - P(\text{GPa})) \ln K_D + 1.828} - 273.15 \quad (23)$$

Applying the multiple regression analysis technique on a large experimental data set and natural high-Mn granulites, Krogh–Ravna (2000) proposed a new garnet–clinopyroxene geothermometer (see row 33 in Tab. 5):

$$T_{KR00}^{Grt-Cpx} (^{\circ}C) = \frac{1939.9 + 3270X_{Ca}^{Grt} - 1396(X_{Ca}^{Grt})^2 + 3319X_{Mn}^{Grt}}{1.223 + \ln K_D} - \frac{3535X_{Mn}^{Grt} + 1105X_{Mg\#}^{Grt} - 3561(X_{Mg\#}^{Grt})^2 + 2324(X_{Mg\#}^{Grt})^3}{1.223 + \ln K_D} + \frac{+169.4P(\text{GPa})}{1.223 + \ln K_D} - 273.15 \quad (24)$$

Based on the compilation of experimental data, the author stressed that the amount of jadeite component in clinopyroxene does not seem to affect the Fe–Mg distribution coefficient between garnet and clinopyroxene. In addition to a significant dependence between the distribution coefficient K_D , X_{Ca}^{Grt} and $X_{Mg\#}^{Grt}$, Krogh–Ravna's (2000) empirical formulation also considers the effect of X_{Mn}^{Grt} in temperature estimation.

Over 300 garnet–clinopyroxene pairs from published experimental data were used by Nakamura (2009) to develop a new formulation of the garnet–clinopyroxene geothermometer (see row 34 in Tab. 5) based on statistical analysis technique:

$$T_{N09}^{Grt-Cpx} (^{\circ}C) = \frac{A - B - C}{D} - 273.15 \quad (25)$$

where

$$A = 2784 + 14.52P(\text{kbar}) + (2601 + 1.44P(\text{kbar})) \times (2X_{Ca}^{Grt}X_{Mg}^{Grt} - a) + (1183 + 6.98P(\text{kbar}))((X_{Ca}^{Grt})^2 - a),$$

$$B = 105(2X_{Ca}^{Grt}X_{Fe}^{Grt} + b) + (814.6 + 3.61P(\text{kbar})) \times ((X_{Ca}^{Grt})^2 + b) - (254.6 + 8.42P(\text{kbar}))(2X_{Mg}^{Grt}X_{Fe}^{Grt} - (X_{Fe}^{Grt})^2 + c),$$

$$C = 83.6((X_{Mg}^{Grt})^2 - 2X_{Mg}^{Grt}X_{Fe}^{Grt} + c) + 1388X_{Mn}^{Grt} - 462(X_{Mg}^{Grt} - X_{Fe}^{Grt}),$$

$$D = \ln K_D + 1.431 + 0.695(2X_{Ca}^{Grt}X_{Mg}^{Grt} + (X_{Ca}^{Grt})^2 - 2a) + 0.203((X_{Ca}^{Grt})^2 - 2X_{Ca}^{Grt}X_{Fe}^{Grt}) + 0.922X_{Mn}^{Grt},$$

$$a = 0.5X_{Ca}^{Grt}(X_{Mg}^{Grt} - X_{Fe}^{Grt} - X_{Mn}^{Grt}), b = 0.5X_{Ca}^{Grt}(X_{Mg}^{Grt} - X_{Fe}^{Grt} + X_{Mn}^{Grt}) \text{ and } c = 0.5(X_{Ca}^{Grt} + X_{Mn}^{Grt})(X_{Mg}^{Grt} - X_{Fe}^{Grt}).$$

According to Nakamura (2009), tests of his garnet–clinopyroxene geothermometer for the accumulated data set are concordant with the experimental temperatures over the whole range of the experimental temperatures be-

tween 800 and 1820 °C, with a standard deviation of 74 °C.

3.1.3. Garnet–orthopyroxene geothermometers

Compared to the garnet–clinopyroxene geothermometry, the temperature dependence of the Fe²⁺–Mg distribution between garnet and orthopyroxene is less pronounced, but recognized as a potential indicator of the *P*–*T* conditions of formation for a variety of natural assemblages,

Tab. 5 Garnet–clinopyroxene geothermometer estimations by WinGrt program

Row		SG1	SG2	SG3	SG4	SG5
1	SiO ₂	40.230	39.920	40.090	40.060	40.480
2	TiO ₂	0.550	0.760	0.520	0.660	0.530
3	Al ₂ O ₃	21.870	21.450	22.010	21.820	22.380
4	FeO	19.030	19.170	18.480	18.590	16.000
5	MgO	10.070	9.650	10.340	10.620	12.700
6	MnO	0.400	0.450	0.410	0.460	0.410
7	CaO	7.270	8.090	7.740	7.560	7.350
8	Na ₂ O	0.000	0.000	0.000	0.000	0.000
9	∑ (wt. %)	99.420	99.490	99.590	99.770	99.850
Garnet recalculation based on 8 cations (<i>apfu</i>)						
10	Si	3.036	3.025	3.018	3.013	3.001
11	Ti	0.031	0.043	0.029	0.037	0.030
12	Al	1.945	1.916	1.953	1.934	1.956
13	Cr	0.000	0.000	0.000	0.000	0.000
14	Fe ²⁺	1.201	1.215	1.164	1.169	0.992
15	Mg	1.133	1.090	1.161	1.191	1.404
16	Mn	0.026	0.029	0.026	0.029	0.026
17	Ca	0.588	0.657	0.624	0.609	0.584
18	Na	0.000	0.000	0.000	0.000	0.000
19	∑ cations	7.960	7.974	7.976	7.983	7.992
Garnet species		Alm	Alm	Alm	Prp	Prp
Input clinopyroxene cations (<i>apfu</i>) and pressure (kbar) for geothermometer estimations						
20	Al _{tot}	0.437	0.468	0.465	0.376	0.388
21	Fe _{tot}	0.191	0.221	0.228	0.216	0.202
22	Fe ²⁺	0.191	0.221	0.228	0.216	0.202
23	Mg	0.536	0.507	0.538	0.593	0.664
24	<i>P</i> (kbar)	25	25	25	25	25
Garnet–clinopyroxene geothermometers (°C)						
25	<i>T</i> 1	1012	1071	1104	1050	1112
26	<i>T</i> 2	1037	1129	1157	1085	1149
27	<i>T</i> 3	1072	1142	1171	1112	1172
28	<i>T</i> 4	1025	1122	1152	1076	1143
29	<i>T</i> 5	1026	1146	1178	1086	1160
30	<i>T</i> 6	1007	1137	1170	1067	1129
31	<i>T</i> 7	1077	1144	1174	1118	1182
32	<i>T</i> 8	1003	1123	1152	1060	1115
33	<i>T</i> 9	1087	1183	1221	1140	1218
34	<i>T</i> 10	1055	1126	1165	1101	1175

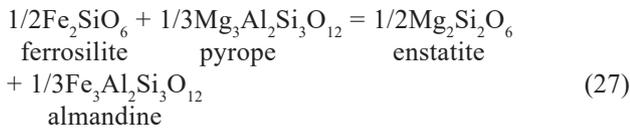
Notes: The formulae were recalculated to 12 anions and 8 cations; sample numbers SG1 to SG5 from Nakamura and Hirajima (2005); Alm = almandine, Prp = pyrope; Garnet–clinopyroxene geothermometer of *T*1 by Råheim and Green (1974), *T*2 by Ellis and Green (1979), *T*3 by Ganguly (1979), *T*4 by Powell (1985), *T*5 by Krogh (1988), *T*6 by Ai (1994), *T*7 by Ganguly et al. (1996), *T*8 by Liu (1998), *T*9 by Krogh–Ravna (2000), *T*10 by Nakamura (2009).

especially for basic granulites and charnockites that formed at granulite-facies metamorphic conditions in the Earth's upper mantle. Garnet–orthopyroxene pairs coexist at 700–900 °C (rarely up to 1000 °C) and 4–10 kbar in rocks including charnockites, basic granulites, aluminous granulites, pyroxene hornfelses and metamorphic iron formations. Coexisting garnet–orthopyroxene pairs are also found in mantle-derived lherzolites, eclogites and peridotites at > 800 °C and 10 kbar (Lal 1993).

Dahl (1980) studied the garnet–clinopyroxene and garnet–orthopyroxene compositions in diverse metamorphic lithologies from two small areas in the Ruby Range, southwestern Montana, and proposed an empirical calibration of garnet–orthopyroxene geothermometer (see row 27 in Tab. 6) based on a limited mineral-pair analysis by using the multiple linear regression technique:

$$T_{D80}^{\text{Grt-Opx}} (\text{°C}) = \frac{1391 + 1509(X_{\text{Fe}}^{\text{Grt}} - X_{\text{Mg}}^{\text{Grt}}) + 2810X_{\text{Ca}}^{\text{Grt}}}{1.987 \ln K_D} + \frac{2855X_{\text{Mn}}^{\text{Grt}}}{1.987 \ln K_D} - 273.15 \quad (26)$$

where K_D is $(\text{Fe}^{2+}/\text{Mg})^{\text{Grt}}/(\text{Fe}^{2+}/\text{Mg})^{\text{Opx}}$ for the following garnet–orthopyroxene Fe^{2+} –Mg exchange reaction:



Neglecting the non-ideality of the Fe–Mg substitution in garnet and orthopyroxene phases and the effect of pressure, Raith et al. (1983) developed an empirical garnet–orthopyroxene geothermometer (see row 28 in Tab. 6) using the compositional data from garnet lherzolite nodules in the granulite-facies terrain of South India:

$$T_{R83}^{\text{Grt-Opx}} (\text{°C}) = \frac{1684}{0.334 + \ln K_D} - 273.15 \quad (28)$$

The partitioning of Fe and Mg between garnet and aluminous orthopyroxene was experimentally investigated by Harley (1984) within the P – T range of 5–30 kbar and 800–1200 °C in the FeO – MgO – Al_2O_3 – SiO_2 and CaO – FeO – MgO – Al_2O_3 – SiO_2 systems. The calibration of garnet–orthopyroxene geothermometer (see row 29 in Tab. 6) is applicable to garnet peridotites and granulites:

$$T_{H84}^{\text{Grt-Opx}} (\text{°C}) = \frac{3740 + 1400X_{\text{Ca}}^{\text{Grt}} + 22.86P(\text{kbar})}{1.9872 \ln K_D + 1.96} - 273.15 \quad (29)$$

Although the accuracy and precision of Harley's (1984) calibration are limited due to large relative errors in the experimental and natural-rock data, it was however pointed out that this geothermometer yields reasonable temperature estimates for a variety of natural samples.

Taking into account the enthalpy, entropy and the volume data of Mg and Fe^{2+} end-member components in garnet and orthopyroxene solid solutions, a new Fe^{2+} –Mg exchange geothermometer (see row 30 in Tab. 6) for coexisting orthopyroxene–garnet pairs was formulated by Sen and Bhattacharya (1984) for granulite-facies quartzofeldspathic and basic lithologies:

$$T_{SB84}^{\text{Grt-Opx}} (\text{°C}) = \frac{2713 + 22P(\text{kbar}) + 3300X_{\text{Ca}}^{\text{Grt}} + 195(X_{\text{Fe}}^{\text{Grt}} - X_{\text{Mg}}^{\text{Grt}})}{-1.9872 \ln K_D + 0.787 + 1.5X_{\text{Ca}}^{\text{Grt}}} - 273.15 \quad (30)$$

where $K_D = (\text{Fe}^{2+}/\text{Mg})^{\text{Opx}}/(\text{Fe}^{2+}/\text{Mg})^{\text{Grt}}$.

Based on an experimental study for garnet and orthopyroxene pairs, Lee and Ganguly (1984) derived an exchange Fe^{2+} –Mg geothermometer (see row 31 in Tab. 6) with an empirical adjustment from natural data set in terms of Ca and Mn contents:

$$T_{LG84}^{\text{Grt-Opx}} (\text{°C}) = \frac{2187 + 1510(X_{\text{Ca}}^{\text{Grt}} - X_{\text{Mn}}^{\text{Grt}}) + 8.6P(\text{kbar})}{\ln K_D + 1.071} - 273.15 \quad (31)$$

Using thermodynamic parameters listed in their table and corresponding activity expressions, the following equation of the garnet–orthopyroxene geothermometer (Eq. 32; see row 32 in Tab. 6) has been derived by Perchuk et al. (1985) to estimate the physical and geodynamic conditions of metamorphism of the Precambrian granulites of the Aldan Shield, eastern Siberia, on the basis of analytical, petrological and experimental data:

$$T_{P85}^{\text{Grt-Opx}} (\text{°C}) = \frac{4766 + 2533(X_{\text{Fs}}^{\text{Opx}} - X_{\text{En}}^{\text{Opx}}) - 5214X_{\text{Al}}^{\text{Opx}}}{1.9872 \ln K_D + 2.65 + 1.86(X_{\text{Fs}}^{\text{Opx}} - X_{\text{En}}^{\text{Opx}}) + 1.242X_{\text{Ca}}^{\text{Grt}}} + \frac{5704X_{\text{Ca}}^{\text{Grt}} + 23P(\text{kbar})}{1.9872 \ln K_D + 2.65 + 1.86(X_{\text{Fs}}^{\text{Opx}} - X_{\text{En}}^{\text{Opx}}) + 1.242X_{\text{Ca}}^{\text{Grt}}} - 273.15 \quad (32)$$

where $K_D = (X_{\text{Fe}}^{\text{Grt}}/X_{\text{En}}^{\text{Opx}})/(X_{\text{Mg}}^{\text{Grt}}/X_{\text{Fs}}^{\text{Opx}})$, $X_{\text{Fs}}^{\text{Opx}} = \text{Fe}/(\text{Fe} + \text{Mg} + 0.5\text{Al})$, $X_{\text{En}}^{\text{Opx}} = \text{Mg}/(\text{Fe} + \text{Mg} + 0.5\text{Al})$ and $X_{\text{Al}}^{\text{Opx}} = 1 - X_{\text{Fs}}^{\text{Opx}} - X_{\text{En}}^{\text{Opx}}$.

Lee and Ganguly (1988) carried out an experimental study to determine the Fe–Mg fractionation between coexisting garnet and orthopyroxene at 20–45 kbar, 975–1400 °C, and the effect of iron on alumina solubility in orthopyroxene at 25 kbar, 1200 °C, and 20 kbar, 975 °C in the FeO – MgO – Al_2O_3 – SiO_2 system. Using their own experimental data, Lee and Ganguly (1988) formulated a revised version of their previous (i.e. Lee and Ganguly 1984) garnet–orthopyroxene exchange geothermometer (see row 33 in Tab. 6), applicable to a wide variety of natural garnet–orthopyroxene assemblages:

Tab. 6 Garnet–orthopyroxene geothermometer estimations by WinGrt program

Row		SG1	SG2	SG3	SG4
1	SiO ₂	38.500	38.150	37.800	38.100
2	TiO ₂	0.000	0.030	0.050	0.060
3	Al ₂ O ₃	21.210	21.840	21.460	21.740
4	Cr ₂ O ₃	0.010	0.120	0.000	0.120
5	FeO	30.940	29.980	33.790	29.740
6	MgO	6.850	6.830	4.120	7.620
7	MnO	0.720	0.890	1.280	0.640
8	CaO	1.770	2.760	2.890	1.890
9	Na ₂ O	0.030	0.020	0.030	0.040
10	∑ (wt. %)	100.030	100.620	101.420	99.950
Garnet recalculation based on 8 cations (<i>apfu</i>)					
11	Si	3.015	2.963	2.971	2.967
12	Ti	0.000	0.002	0.003	0.004
13	Al	1.958	1.999	1.988	1.995
14	Cr	0.001	0.007	0.000	0.007
15	Fe ³⁺	0.016	0.068	0.068	0.063
16	Fe ²⁺	2.010	1.880	2.154	1.874
17	Mg	0.800	0.791	0.483	0.885
18	Mn	0.048	0.059	0.085	0.042
19	Ca	0.149	0.230	0.243	0.158
20	Na	0.005	0.003	0.005	0.006
21	∑ cations	8.000	8.000	8.000	8.000
Garnet species					
		Alm	Alm	Alm	Alm
Input orthopyroxene cations (<i>apfu</i>) and pressure (kbar) for geothermometer estimations					
22	Al _{tot}	0.176	0.114	0.107	0.115
23	Fe _{tot}	0.891	0.906	1.173	0.858
24	Fe ²⁺	0.891	0.906	1.173	0.858
25	Mg	0.974	1.008	0.772	1.097
26	<i>P</i> (kbar)	7	7	7	7
Garnet–orthopyroxene geothermometers (°C)					
27	<i>T</i> 1	814	876	920	781
28	<i>T</i> 2	979	1016	920	993
29	<i>T</i> 3	727	757	707	736
30	<i>T</i> 4	810	854	792	819
31	<i>T</i> 5	831	869	811	843
32	<i>T</i> 6	776	837	822	795
33	<i>T</i> 7	795	835	775	807
34	<i>T</i> 8	713	764	744	734
35	<i>T</i> 9	788	822	789	794
36	<i>T</i> 10	801	850	810	817

Notes: The formulae were recalculated to 12 anions and 8 cations; Fe³⁺ and Fe²⁺ estimations from total FeO (wt. %) contents were carried out on the basis of stoichiometric constraints using Droop's (1987) method; sample numbers SG 1 to SG4 from Sen and Bhattacharya (1984); Alm = almandine; Geothermometers *T*1 by Dahl (1980), *T*2 by Raith et al. (1983), *T*3 by Harley (1984), *T*4 by Sen and Bhattacharya (1984), *T*5 by Lee and Ganguly (1984), *T*6 by Perchuk et al. (1985), *T*7 by Lee and Ganguly (1988), *T*8 by Perchuk and Lavrent'eva (1990), *T*9 by Bhattacharya et al. (1991), *T*10 by Lal (1993)

$$T_{\text{LG88}}^{\text{Grt-OpX}} (\text{°C}) = \frac{1981 + 1509.66(X_{\text{Ca}}^{\text{Grt}} - X_{\text{Mn}}^{\text{Grt}}) + 11.91P(\text{kbar})}{\ln K_D + 0.97} - 273.15 \quad (33)$$

Based on the experimental data at temperatures of 700 and 800 °C over a wide range of pressures (6–30

kbar) for the orthopyroxene–garnet Fe²⁺–Mg exchange reaction, Perchuk and Lavrent'eva (1990) derived a new garnet–orthopyroxene geothermometer (see row 34 in Tab. 6) in the third iteration of the non-ideal nature of the Fe–Mg–Al solid solution in orthopyroxene, as well as taking into account the variable content of Ca in garnet and Al in orthopyroxene:

$$T_{PL90}^{Grt-Opx} (^{\circ}C) = \frac{4066 - 347(X_{En}^{Opx} - X_{Fs}^{Opx}) - 17484X_{Al}^{Opx}}{A} + \frac{5769X_{Ca}^{Grt} + 23.42P(\text{kbar})}{A} - 273.15 \quad (34)$$

where $A = 1.987 \ln K_D + 2.143 + 0.0929(X_{En}^{Opx} - X_{Fs}^{Opx}) - 12.8994X_{Al}^{Opx} + 3.846X_{Ca}^{Grt}$.

Using the retrieved data combined with the known thermochemical parameters for the pyrope–grossular and almandine–grossular binaries, Bhattacharya et al. (1991) published a refined formulation of the garnet–orthopyroxene geothermometer (see row 35 in Tab. 6) based exclusively on calorimetric determinations and high-quality phase-equilibrium data:

$$T_{B91}^{Grt-Opx} (^{\circ}C) = \frac{1611 + 21P(\text{kbar}) + 906X_{Ca}^{Grt} + A}{\ln K_D + 0.796} + \frac{477(2X_{En}^{Opx} - 1)}{\ln K_D + 0.796} - 273.15 \quad (35)$$

where $A = -1220X_{Fe}^{Grt} X_{Mg}^{Grt} X_{Ca}^{Grt} (X_{Mg}^{Grt} - X_{Fe}^{Grt}) - 136(X_{Mg}^{Grt})^2 + 746(X_{Fe}^{Grt})^2$.

According to Bhattacharya et al. (1991), Eq. 35 can be used to estimate the temperatures from garnet–orthopyroxene pairs for $X_{Ca}^{Grt} \leq 30$ mol %, $X_{Mn}^{Grt} \leq 0.05$ mol % and pressures < 15 kbar.

Lal (1993) refined the calibration of the Fe²⁺–Mg exchange garnet–orthopyroxene geothermometer (see row 36 in Tab. 6) using the large dataset for the FeO–MgO–Al₂O₃–SiO₂ and CaO–FeO–MgO–Al₂O₃–SiO₂–H₂O systems in a wide range of pressure and temperature conditions (700–1400 °C and 5–50 kbar):

$$T_{L93}^{Grt-Opx} (^{\circ}C) = \frac{3367 + 21P - 1(\text{kbar})24A - 948(X_{Fs}^{Opx} - X_{En}^{Opx})}{1.987 \ln K_D + 1.634 + B - 0.34(X_{Fs}^{Opx} - X_{En}^{Opx})} - \frac{1950X_{Al}^{Opx}}{1.987 \ln K_D + 1.634 + B - 0.34(X_{Fs}^{Opx} - X_{En}^{Opx})} - 273.15 \quad (36)$$

where

$$A = -1256X_{Mg}^{Grt} - 2880(X_{Fe}^{Grt})^2 + 8272X_{Mg}^{Grt}X_{Fe}^{Grt} + 812X_{Ca}^{Grt} \times (X_{Mg}^{Grt} - X_{Fe}^{Grt}) + 90(X_{Ca}^{Grt})^2 - 2340X_{Ca}^{Grt}(X_{Fe}^{Grt} + X_{Mg}^{Grt}) - 3047X_{Mg}^{Grt}X_{Ca}^{Grt} - 1813X_{Fe}^{Grt}X_{Ca}^{Grt} - 4498X_{Ca}^{Grt}(X_{Fe}^{Grt} - X_{Mg}^{Grt})$$

and

$$B = (X_{Mg}^{Grt})^2 + 1.7(X_{Fe}^{Grt})^2 - 5.4X_{Fe}^{Grt}X_{Mg}^{Grt} - 0.35X_{Ca}^{Grt} \times (X_{Mg}^{Grt} - X_{Fe}^{Grt}) + 1.5(X_{Ca}^{Grt})^2 + 1.666X_{Ca}^{Grt}(X_{Fe}^{Grt} + X_{Mg}^{Grt}) + 0.332X_{Fe}^{Grt}X_{Ca}^{Grt} + 1.516X_{Ca}^{Grt}(X_{Fe}^{Grt} - X_{Mg}^{Grt}).$$

According to Lal (1993), Eq. 36 is applicable to the garnet–orthopyroxene pairs from crustal rocks including

garnet with low Mn content ($X_{Mn}^{Grt} < 5$ mol %), as well as mantle-derived rocks with a low Cr₂O₃ content in garnet (< 5 wt. %).

3.2. Garnet in provenance studies

Parallel to the development of microanalysis equipment, a wide group of detrital minerals such as pyroxene, amphibole, epidote, tourmaline, and garnet were considered as potential provenance indicators in sedimentary environments. Although heavy minerals have been used in a variety of provenance studies, some of them are sensitive to diagenesis and many species suffer from the dissolution process known as intrastratal solution (Morton 1985). For example pyroxene, amphibole and epidote group minerals are moderately or extremely susceptible to intrastratal solution under both deep burial and weathering conditions. Even though composition of tourmaline, as a refractory mineral, may be used in provenance studies (e.g. Henry and Dutrow 2018), garnet minerals remain the most suitable group to determine the provenance of sediments (Morton 1985).

Because garnet composition is a function of P – T conditions, and garnet is comparatively stable during surface weathering, transport, and deep burial, several geochemical discrimination schemes have been developed and applied as a tool in sedimentary provenance analysis (Schönig et al. 2018). Wright (1938) made the first attempt to classify garnets from different host-rock types in ternary (almandine + spessartine)–pyrope–grossular (mol. %) discrimination diagram (Fig. 5a), including granites and granite pegmatites, biotite schists, amphibolites and eclogites. However, there is a strong overlap between garnets from biotite schists and amphibolites, and between amphibolites and eclogites (Preston et al. 2002; Mange and Morton 2007). The first insight into the potential of garnet geochemistry as a provenance indicator by using electron-microprobe analysis provided the study of Middle Jurassic sandstones from the North Sea by Morton (1985). Ternary garnet classification diagram (Fig. 5b) using the molecular proportions of pyrope–spessartine–grossular, but without involving the Fe-component, was proposed by Teraoka et al. (1997, 1998) for amphibolites, granulites, and eclogite-facies rocks. Similarly, Morton et al. (2004) used the (almandine + spessartine)–pyrope–grossular projection (Fig. 5c) of garnet compositions from Scottish and Norwegian basement terrains, but with the newly defined A, B and C fields, and suggested that garnet compositions can be directly tied back to the adjacent source areas. Modified version of this diagram (Fig. 5d) with an additional D field was introduced by Mange and Morton (2007). Aubrecht et al. (2009) proposed two new ternary classification diagrams (Fig. 5e–f) using molecular proportions of almandine–pyrope–grossular for (ultra-)

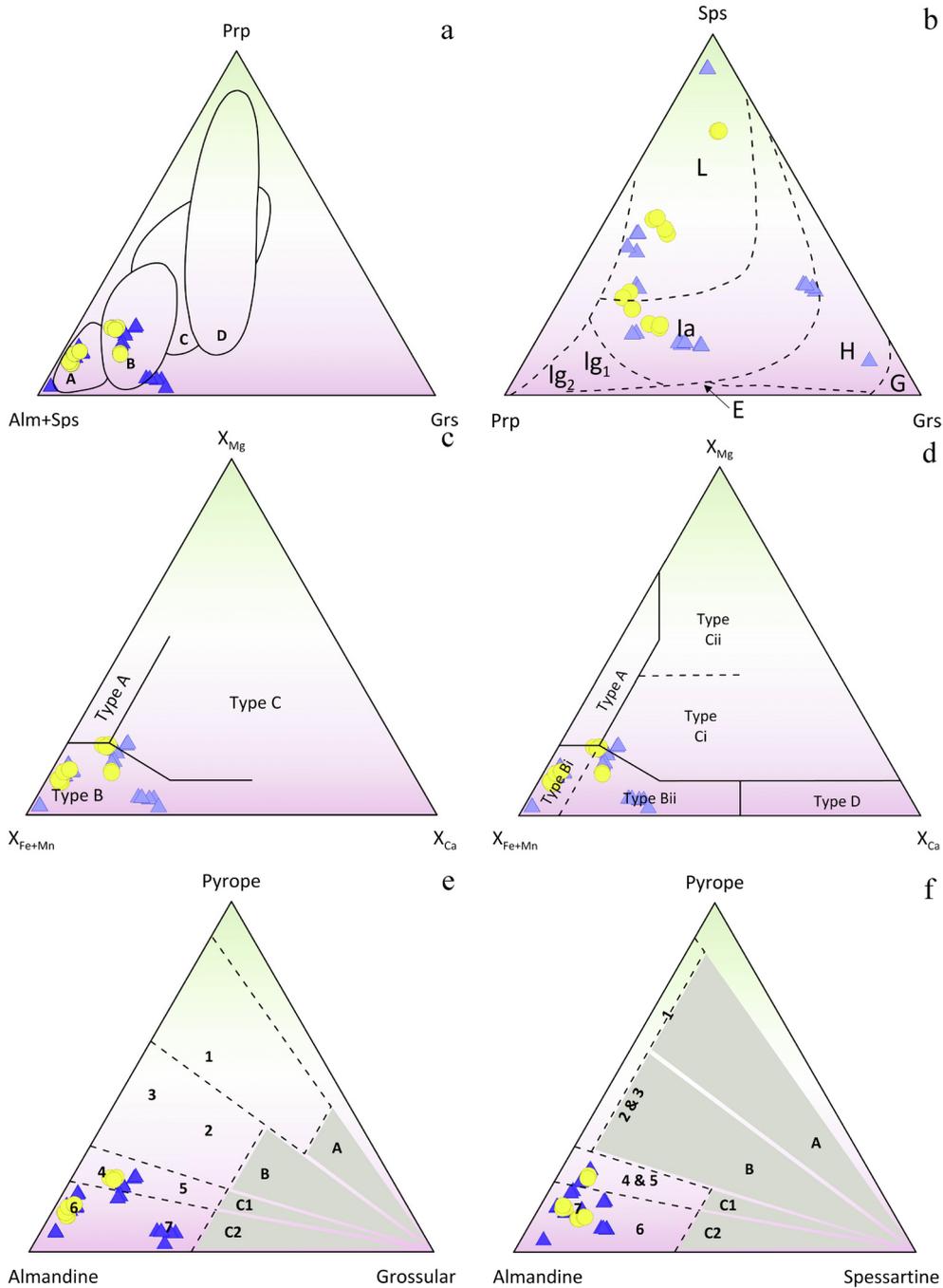


Fig. 5 Ternary discrimination diagrams using molar fractions (%) of garnet end-members in provenance study of sediments. **a** – (Almandine + spessartine)–pyrope–grossular. A = granites and granite pegmatites, B = biotite schists, C = amphibolites, D = eclogites (Wright 1938). **b** – Pyrope–spessartine–grossular. L = low P - T , la = intermediate P - T (up to amphibolite facies), H = high P - T , lg₁, lg₂ = intermediate P - T (granulite facies), E = eclogite, G = grandite garnets (Teraoka et al. 1997, 1998). **c** – $(X_{Fe} + X_{Mn}) - X_{Mg} - X_{Ca}$. A = low Ca, high Mg, B = low Mg, variable Ca, C = high Ca, high Mg (Morton et al. 2004). **d** – $(X_{Fe} + X_{Mn}) - X_{Mg} - X_{Ca}$. Garnets from: A = mainly high-grade granulite-facies metasediments or charnockites and intermediate felsic igneous rocks, B = amphibolite-facies metasedimentary rocks, Bi = intermediate to felsic igneous rocks, Ci = mainly high-grade mafic rocks, Cii = ultramafics with high Mg (pyroxenites and peridotites), D = metasomatic rocks, very low-grade metabasic rocks and ultrahigh-temperature calc-silicate granulites (Mange and Morton 2007). **e** – Almandine–pyrope–grossular and **f** – Almandine–pyrope–spessartine. Garnets from: A = high-pressure (HP) to ultrahigh-pressure (UHP) rocks, B = eclogite- and granulite-facies rocks, C1 = higher amphibolite- to granulite-facies rocks, C2 = amphibolite-facies rocks but also any other rocks such as blueschists, skarns, serpentinites and igneous rocks. Garnets derived from: 1 = UHP eclogites or garnet peridotites, 2 = HP eclogites and HP mafic granulites, 3 = felsic–intermediate granulites, 4 = gneisses metamorphosed under P - T conditions transitional between granulite- and amphibolite-facies metamorphism, 5 = amphibolites metamorphosed under P - T conditions transitional between granulite- and amphibolite-facies metamorphism, 6 = amphibolite-facies gneisses, 7 = amphibolites proper (Aubrecht et al. 2009). Garnet analyses in all plots are taken from Suggate and Hall (2014). Triangles and circles correspond to tables 4 and 5 in Suggate and Hall (2014), respectively.

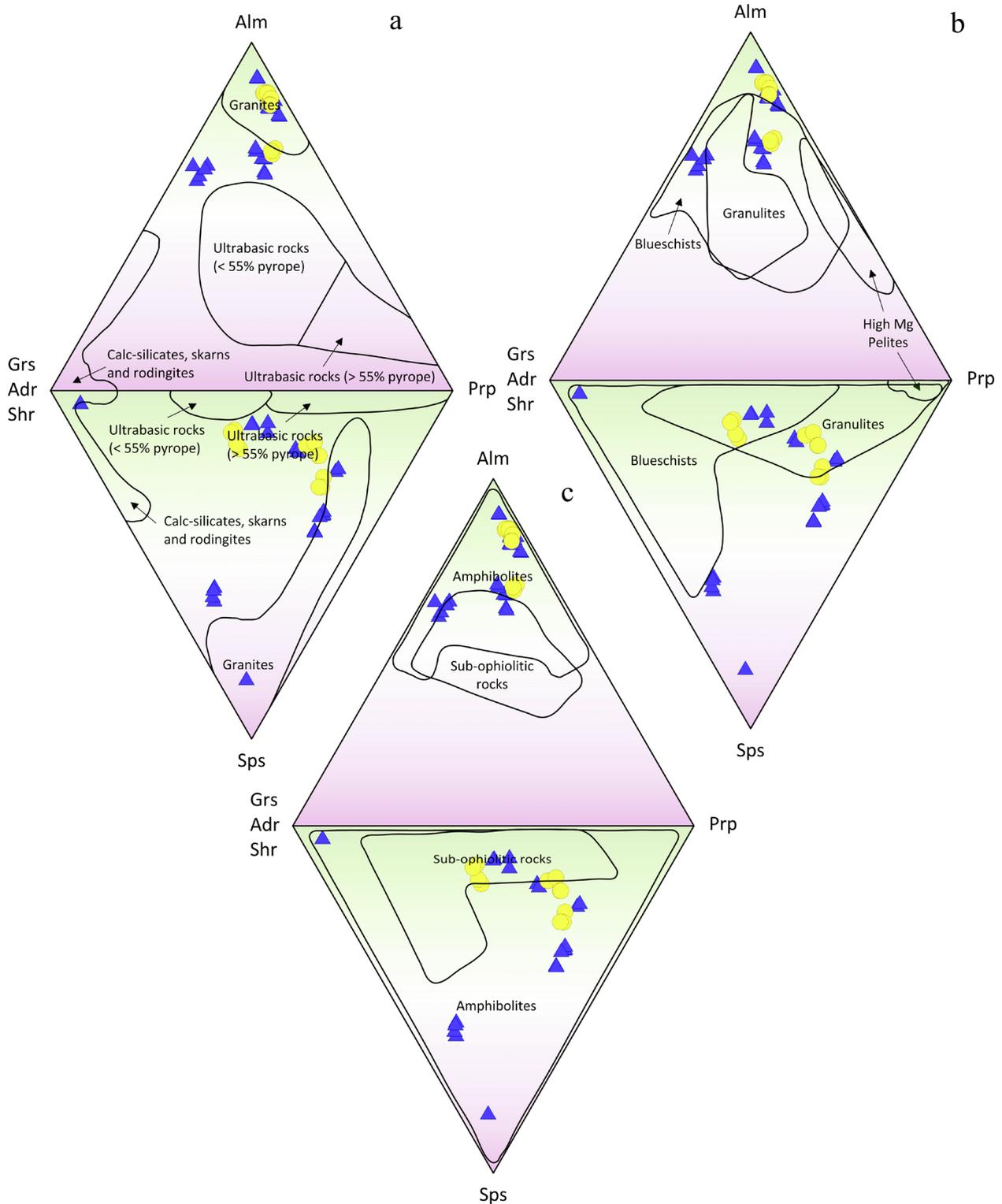


Fig. 6 Ternary plots based on garnet end-members discriminating garnets with different protoliths (Suggate and Hall 2014). **a** – Ultramafic rocks (peridotites, eclogites and kimberlites), granites, calc-silicates, skarns and rodingites. **b** – Granulites, granulite-facies high-Mg pelites, and blueschists. **c** – Amphibolites and metabasic sub-ophiolitic rocks. Triangles and circles correspond to tables 4 and 5 in Suggate and Hall (2014), respectively.

high-pressure metamorphic rocks and almandine–pyrope–spessartine for granulite- and amphibolite-facies rocks (after Méres 2008).

Suggate and Hall (2014) presented a new approach to garnet discrimination that invokes two ternary diagrams (Fig. 6) with almandine and spessartine apices, and a

common axis (grossular + andradite + schorlomite)–pyrope. Their scheme requires a stepwise separation of garnets of specific composition. Hence, garnets with an uncommon content of Y_2O_3 , V_2O_5 , and ZrO_2 are first removed and then those with high TiO_2 are separated in their discrimination diagrams. This type of projection enables to group garnets derived from ores, skarns, mafic pyroclastic rocks, and nepheline syenites. In a next step, another group of garnets with high pyrope and uvarovite end-member compositions derived from ultramafic rocks is distinguished. Following the second step, all the remaining garnets, i.e., those from amphibolite-, granulite-, and eclogite-facies rocks, are plotted in ternary diagrams with almandine and spessartine apices. These garnets may overlap, but many mafic eclogites can be distinguished by their low spessartine content (Krippner et al. 2014; Suggate and Hall 2014).

4. Summary and availability of the program

WinGrt is a user-friendly program, which is specially developed for personal computers running in the Windows operating system to estimate and classify the garnet supergroup mineral analyses obtained by electron-microprobe or wet-chemical analyses. The program calculates multiple (up to 300) garnet supergroup analyses on the basis of 8 cations and 12 anions. Following the procedure by Droop (1987), WinGrt estimates the Fe^{2+} and Fe^{3+} and, if necessary, Mn^{3+} (*apfu*) contents from electron-microprobe analyses using the stoichiometric constraints. It also provides users a large number of geothermometers to determine the temperature conditions based on the Fe^{2+} –Mg exchange reactions of garnet–biotite, garnet–clinopyroxene, and garnet–orthopyroxene pairs in greenschist- to eclogite-facies metamorphic rocks. The program yields end-member molecular proportions using various normalization schemes to plot ternary discrimination diagrams in sedimentary provenance analysis by using the Golden Software's Grapher software.

WinGrt generates two main windows. The first (i.e. *Start-up/Data Entry Screen*), with several pull-down menus and equivalent shortcuts, enables to edit garnet analyses (wt. %) and recalculated specific input biotite, clinopyroxene and orthopyroxene cation values such as Fe^{2+} , Mg, Al^{VI} , Ti, Fe_{tot} and Al_{tot} (*apfu*) into the data entry section as well as to carry out necessary arrangements for a desired calculation scheme. By clicking the *Calculate* icon (i.e. Σ) in the *Data Entry Screen*, all calculated parameters are displayed in the second window (i.e. *Calculation Screen*). WinGrt reports the output in a tabulated form with columns numbered from 1 to 187 in the *Calculation Screen* window. Garnet-only-related pa-

rameters are listed at columns 1–140. Calculated garnet–biotite, garnet–clinopyroxene and garnet–orthopyroxene geothermometers with input parameters are displayed by light blue, green and pink colours in columns 142–156, 158–172 and 174–187 of the *Calculation Screen*, as well as in an Output Excel file. The results in the *Calculation Screen* can be exported to Microsoft® Excel file (i.e. *Output.xlsx*), by clicking the *Send Results to Excel File (Output.xlsx)* icon or selecting the *Send Results to Excel File (Output.xlsx)* option from the *Excel* pull-down menu. Subsequently, this file is opened in Excel by clicking the *Open and Edit Excel File (Output.xlsx)* icon or selecting the *Open Excel File (Output.xlsx)* option from the *Excel* pull-down menu.

WinGrt is a compiled program that consists of a self-extracting setup file including all the necessary support files (i.e. *.dll and *.ocx) for the 32-bit system. During the setup procedure, the program and its associated files (i.e. support files, help file, data files with the extensions of gsg, xls, xlsx and plot files with the extension of grf) are installed into the personal computer (i.e. the directory C:\Program Files\WinGrt) with the Windows XP or later operating systems. An installation of program into a personal computer with the 64-bit operating system may require the msflexgrd adjustment. This procedure is explained in detail in Electronic Supplementary Material (ESM 2). The self-extracting setup file is 14 MB and can be obtained from the journal server or from corresponding author on request.

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Electronic supplementary material. Examples of garnet data recalculated and plotted by the WinGrt program, the steps for its execution and self-extracting *WinGrt.exe* setup file are available online at the Journal web site (<http://dx.doi.org/10.3190/jgeosci.303>).

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